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(19) (CA) **APPLICATION FOR CANADIAN PATENT** (12)

(54) Gem-Dialkyl-7-Oxabicycloheptyl Substituted Heterocyclic
Amide Prostaglandin Analogs Useful in the Treatment of
Thrombotic and Vasospastic Disease

(72) Misra, Raj N. - U.S.A. ;

(73) Same as inventor

(30) (US) 799,233 1991/11/27

(57) 30 Claims

Notice: This application is as filed and may therefore contain an
incomplete specification.

Canada

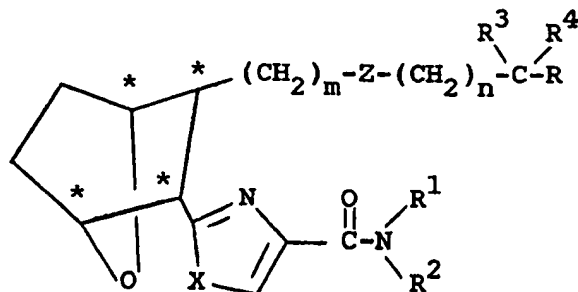
CCA 3254 (10-82) 41 7530-21-936-3254

GEM-DIALKYL-7-OXABICYCLOHEPTYL SUBSTITUTED
HETEROCYCLIC AMIDE PROSTAGLANDIN ANALOGS
USEFUL IN THE TREATMENT OF
THROMBOTIC AND VASOSPASTIC DISEASE

5

The present invention relates to gem-dialkyl-
7-oxabicycloheptyl substituted heterocyclic amide
prostaglandin analogs which are thromboxane A₂
10 (TXA₂) receptor antagonists or combined thromboxane
A₂ receptor antagonists/thromboxane synthetase
inhibitors useful, for example, in the treatment of
thrombotic and/or vasospastic disease. These
compounds have the structural formula I


15 I



20

and including all stereoisomers thereof, wherein

m is 1, 2 or 3; n is 0, 1, 2, 3 or 4;

Z is $-(CH_2)_2-$, $-CH=CH-$ or ,

5 with the proviso that when Z is $-CH=CH-$, n is 1, 2, 3, or 4;

R is CO_2H , CO_2 lower alkyl, or CO_2 alkali metal;

X is O or NH;

10 R^1 is hydrogen, lower alkyl, lower alkenyl, lower alkynyl, aralkyl, aryl, cycloalkyl, cycloalkylalkyl, cycloheteroalkyl, cycloheteroalkylalkyl, heteroaryl or heteroarylalkyl, or amide

15 $(-(CH_2)_t-\overset{\overset{O}{\parallel}}{C}-\underset{\underset{H}{|}}{N}-R_a$ or $-(CH_2)_t-\underset{\underset{H}{|}}{N}-\overset{\overset{O}{\parallel}}{C}-R_a$ wherein t

is 1 to 12 and R_a is lower alkyl, aryl, cycloalkyl, or cycloalkylalkyl), each of R^1 being unsubstituted or optionally substituted with a lower alkyl, aryl, cycloalkyl, or cycloalkylalkyl group;

20 R^2 is hydrogen, lower alkyl, aryl, or aralkyl; or

R^1 and R^2 together with the nitrogen to which they are linked may form a 5- to 8- membered ring;

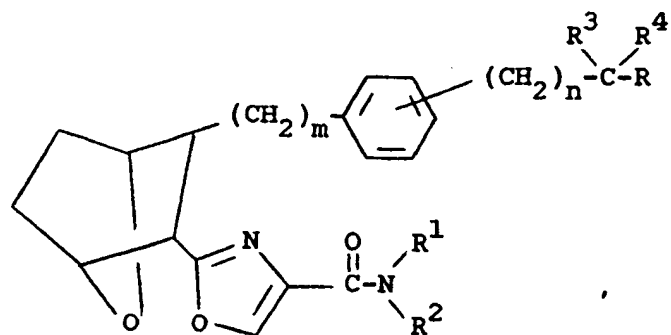
25 and

R^3 and R^4 are the same or different and each is lower alkyl. R^3 and R^4 may be linked to form a 3- or 4-membered ring.

30 Thus, the compounds of the invention include the following types of compounds:

IA

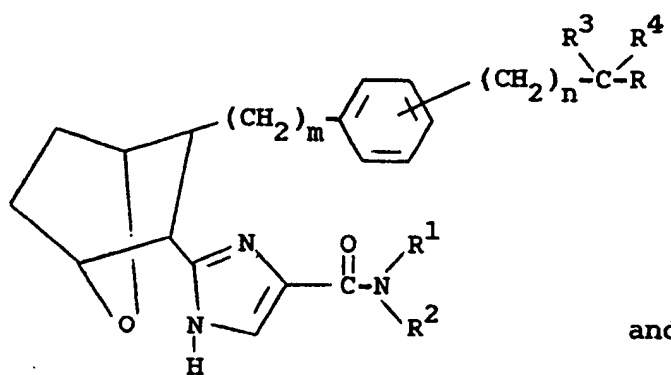
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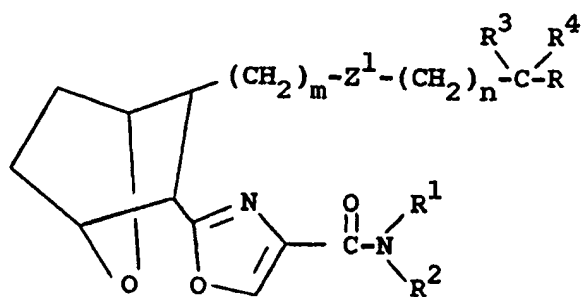


and

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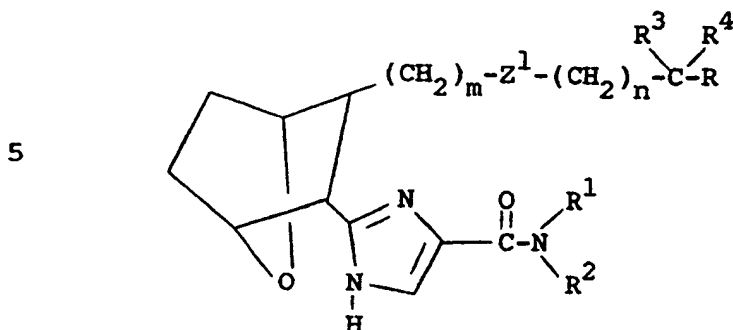
IC

25



, and

ID



10 wherein in formulae IC and ID, Z^1 is $-\text{CH}=\text{CH}-$ or $-(\text{CH}_2)_2-$.

The term "lower alkyl" or "alkyl" as employed herein includes both straight and branched chain radicals of up to 18 carbons, preferably 1 to 12 carbons, such as methyl, ethyl, propyl, isopropyl, butyl, t-butyl, isobutyl, pentyl, hexyl, isohexyl, heptyl, 4,4-dimethylpentyl, octyl, 2,2,4-trimethylpentyl, nonyl, decyl, undecyl, dodecyl, the various branched chain isomers thereof, and the like as well as such groups including 1, 2 or 3 halo substituents, an aryl substituent, an alkyl-aryl substituent, a haloaryl substituent, a cycloalkyl substituent, an alkylcycloalkyl substituent, hydroxy or a carboxy substituent.

25 The term "cycloalkyl" includes saturated cyclic hydrocarbon groups containing 3 to 12 carbons, preferably 3 to 8 carbons, which include cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl, cyclooctyl, cyclodecyl and cyclododecyl, any of which groups may be substituted with substituents such as halogen, lower alkyl, alkoxy and/or hydroxy group.

The term "aryl" or "Ar" as employed herein refers to monocyclic or bicyclic aromatic groups containing from 6 to 10 carbons in the ring portion, such as phenyl, naphthyl. Aryl (or Ar), phenyl or naphthyl may include substituted aryl, substituted phenyl or substituted naphthyl, which may include 1 or 2 substituents on either the phenyl or naphthyl such as lower alkyl, trifluoromethyl, halogen (Cl, Br, I or F), lower alkoxy, arylalkoxy, hydroxy, alkylthio, alkylsulfinyl, alkylsulfonyl, arylthio, arylsulfinyl and/or arylsulfonyl.

The term "aralkyl", "aryl-alkyl" or "aryl-lower alkyl" as used herein refers to lower alkyl groups as discussed above having an aryl substituent, such as benzyl.

The term "lower alkoxy", "alkoxy" or "aralkoxy" includes any of the above lower alkyl, alkyl or aralkyl groups linked to an oxygen atom.

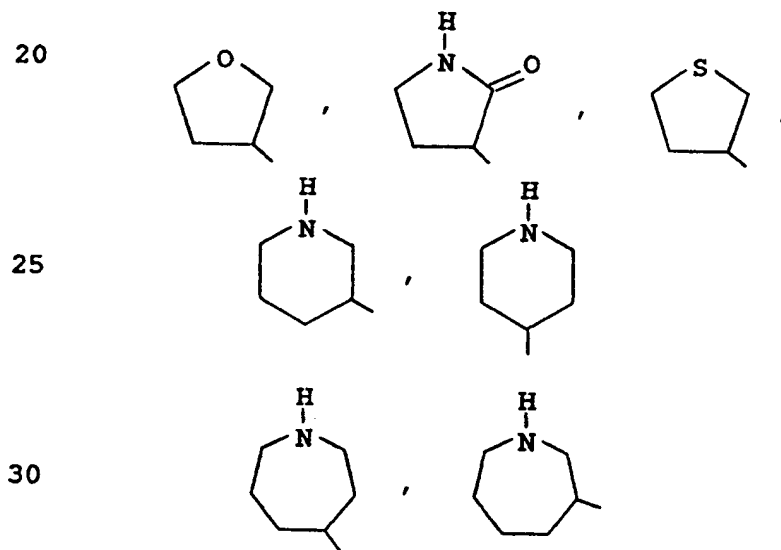
The term "halogen" or "halo" as used herein refers to chlorine, bromine, fluorine or iodine with chlorine being preferred.

The term "lower alkenyl" or "alkenyl" as employed herein with respect to the R^1 substituent includes a carbon chain of up to 16 carbons, preferably 3 to 10 carbons, containing one double bond which will be separated from "N" by at least one saturated carbon moiety such as $-(CH_2)_q-$ where q can be 1 to 14, such as 2-propenyl, 2-butenyl, 3-butenyl, 2-pentenyl, 4-pentenyl and the like, and may include a halogen substituent such as I, Cl, or F.

The term "lower alkynyl" or "alkynyl" as employed herein with respect to the R^1 substituent includes a carbon chain of up to 16 carbons, preferably 3 to 10 carbons, containing one triple bond which will be separated from "N" by at least one saturated carbon moiety such as $-(CH_2)_{q'}-$ where q' can be 1 to 14, such as 2-propynyl, 2-butynyl, 3-butynyl and the like.

The term "cycloheteroalkyl" as used herein as an R^1 substituent refers to a 5-, 6- or 7-membered saturated ring which includes 1 or 2 hetero atoms such as nitrogen, oxygen and/or sulfur, and which

is linked to the "N" of the $-N \begin{smallmatrix} R^1 \\ R^2 \end{smallmatrix}$ group through a carbon atom either beta or gamma to a heteroatom, such as



and the like.

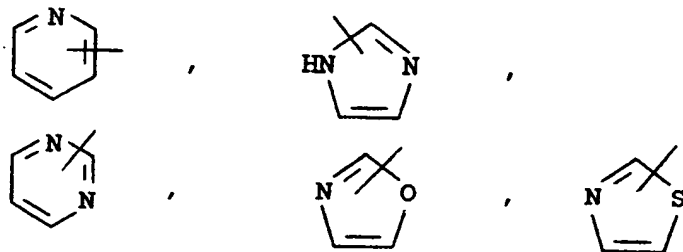
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The term "heteroaryl" or heteroaromatic as
 an R^1 substituent refers to a 5- or 6-membered
 aromatic ring which includes 1 or 2 hetero atoms
 such as nitrogen, oxygen or sulfur, which are not
 5 directly linked through a hetero

atom to the "N" of the $\begin{array}{c} R^1 \\ | \\ -N \\ | \\ R^2 \end{array}$ group,

such as

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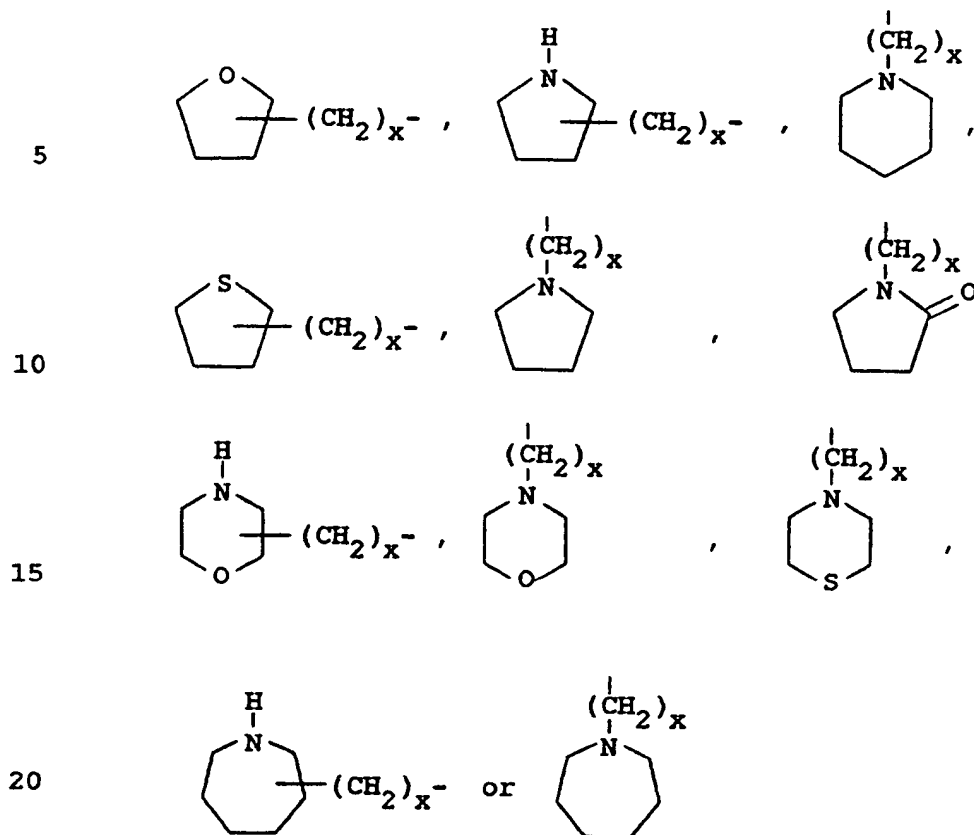
and the like

The term "cycloheteroalkylalkyl" as defined
 20 by R^1 refers to 5-, 6- or 7-membered saturated ring
 which includes 1 or 2 heteroatoms such as nitrogen,
 oxygen or sulfur, and is linked

to the "N" of the $\begin{array}{c} R^1 \\ | \\ -N \\ | \\ R^2 \end{array}$ group through a $(CH_2)_x$

25

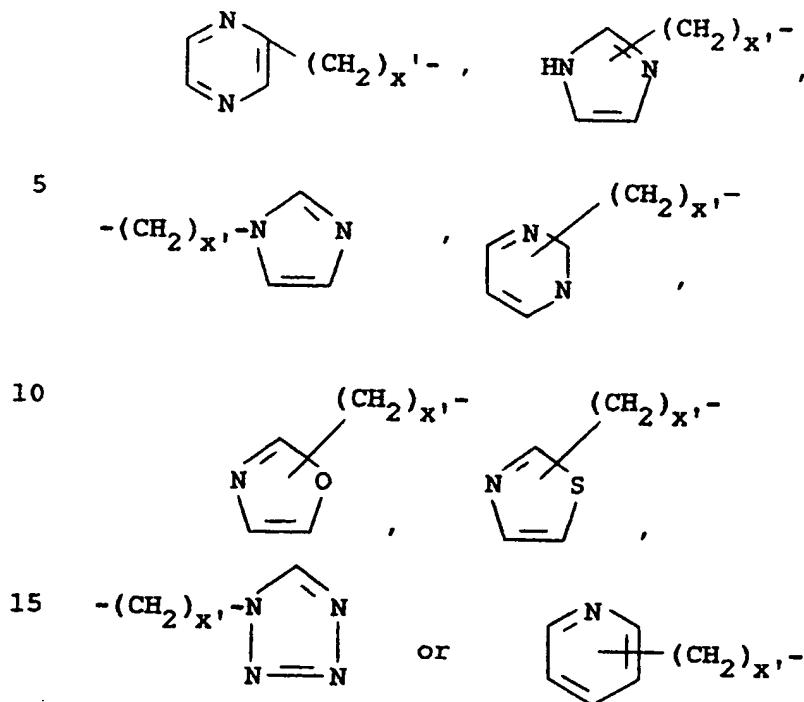
chain wherein x is 1 to 12, preferably 1 to 8,
 such as



The term "heteroarylalkyl" as defined by R^1 refers to a 5-, 6- or 7-membered aromatic ring which includes 1, 2, 3 or 4 heteroatoms such as nitrogen, oxygen or sulfur, and is linked to the

"N" of the $-N \begin{matrix} R^1 \\ R^2 \end{matrix}$ group through a $-(CH_2)_{x'}$ -

chain where x' is 1 to 12, preferably 1 to 8, such as



Preferred are those compounds of formula I

wherein Z is and X is O, and R^3 and R^4 are the same alkyl. More preferred are compounds of formula I wherein R^3 and R^4 are each methyl,

Z- is , m is 1, n is 1 or 2,

X is O, R is CO_2H , R^1 is alkyl or substituted alkyl such as cycloalkylalkyl, such as cyclohexylbutyl and R^2 is H or lower alkyl such as methyl, and


$-(CH_2)_n-\overset{\overset{R^3}{|}}{\underset{\underset{R^4}{|}}{C}}-R$ is in the ortho or meta position.

Also preferred are compounds of formula I wherein Z is $-CH=CH-$ in the cis configuration, m is

1, n is 2 or 3, R is CO₂H, R¹ is alkyl or substituted alkyl such as cycloalkylalkyl and R² is H or lower alkyl, such as methyl, and R³ and R⁴ are each methyl.

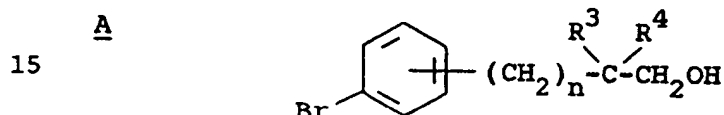
The compounds of formula I of the invention
5 may be prepared as follows.

The various compounds of the invention

wherein Z is 

10 may be prepared as outlined below.

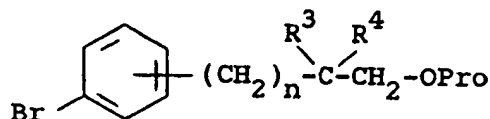
Compounds of the invention where X is O are prepared starting with bromophenylalkyl alcohol A



wherein n is 1, 2, 3 or 4

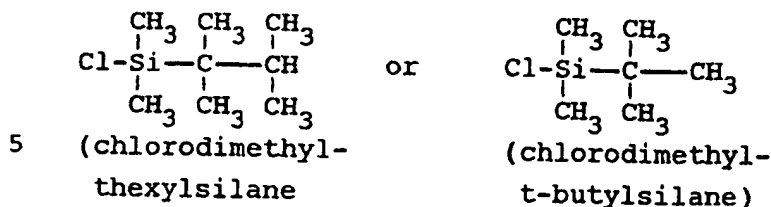
which is treated with a protecting compound such as dimethylhexylsilyl chloride, in the presence of an
20 amine base such as triethylamine or 4-dimethylamino-pyridine, and an inert solvent such as methylene chloride, employing conventional procedures, to form the protected bromophenylalkyl compound B

25 B

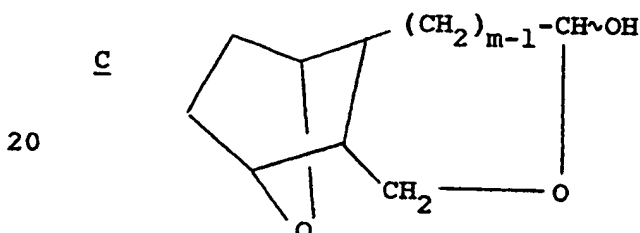


30 wherein Pro represents a protecting group.

Examples of protecting compounds suitable for use herein in reacting with bromophenylalkyl alcohol A include but are not limited to

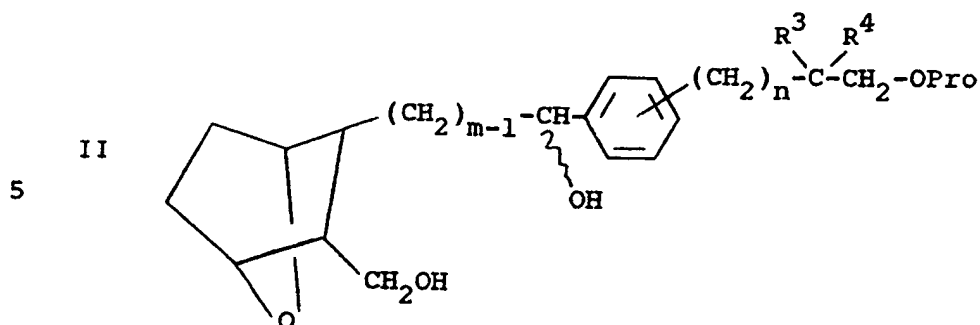


10 The protected compound B is then converted to a Grignard reagent by treatment with magnesium in the presence of an inert organic solvent such as tetrahydrofuran (THF) or diethyl ether and then is condensed with (exo)octahydro-5,8-epoxy-1H-benzo-
 15 pyran-3-ol or (exo)octahydro-4,7-epoxyisobenzofuran-1-ol (prepared as described in U.S. Patent No. 4,143,054) of the structure C



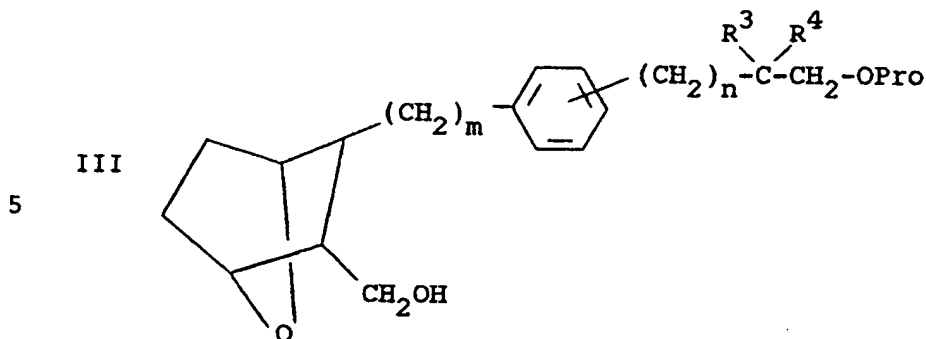
25 employing a molar ratio of C:B of within the range of from about 1:2 to about 1:4, in the presence of an inert organic solvent such as THF at a reduced temperature of from about -78 to about 25°C, to form the condensed 7-oxabicycloheptane compound II

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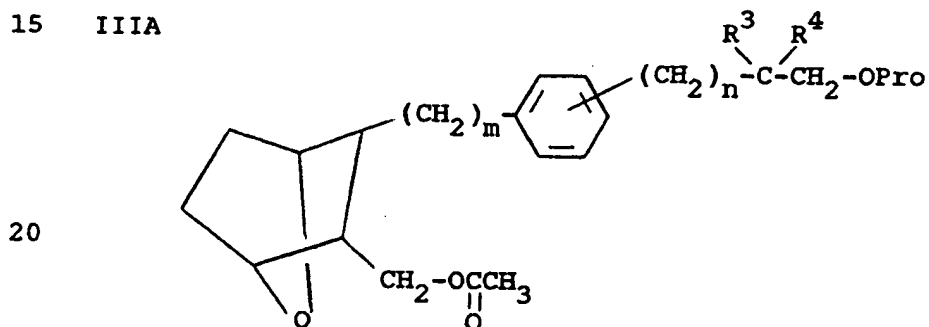


10 In a preferred method, compound II can be
 formed by treatment of C with ethylmagnesium bromide
 employing a molar ratio of C:ethylmagnesium bromide
 of within the range of from about 1:1 to about 1:0.9
 in the presence of an inert organic solvent such as
 15 THF at a reduced temperature of from about -78° to
 about 0°C . Treatment of the resulting anion solu-
 tion with the above Grignard reagent from compound B
 employing a molar ratio of C:Grignard B of within the
 range of 1:1.1 to 1:1.5 at a temperature of about 0°
 20 to about 25°C forms compound II.

The condensed compound II is then subjected
 to hydrogenolysis by treatment with hydrogen in
 the presence of a catalyst such as palladium
 hydroxide on charcoal in acetic acid or an inert
 25 organic solvent such as methanol or ethyl acetate,
 to form the alcohol III



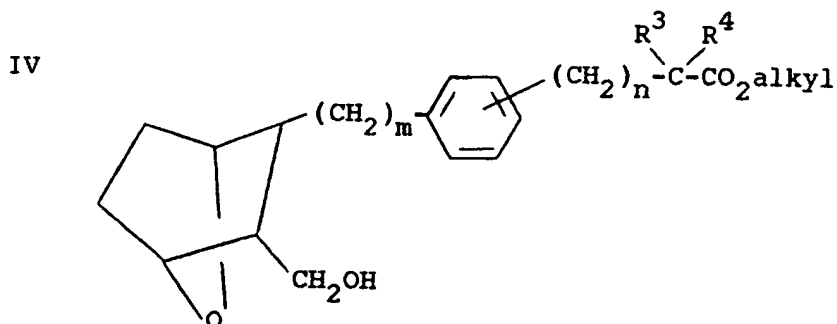
10 Alcohol III is subjected to acetylation by treatment with acetyl chloride or acetic anhydride in the presence of pyridine and methylene chloride to acetylate the free alcohol to form IIIA.



20 The protected alcohol IIIA is then subjected to a Jones oxidation wherein a solution of protected alcohol IIIA in acetone cooled to from about -10 to about 25°C is treated with Jones reagent (that is, CrO₃ dissolved in sulfuric acid in the presence of water, prepared as described in Fieser & Fieser,

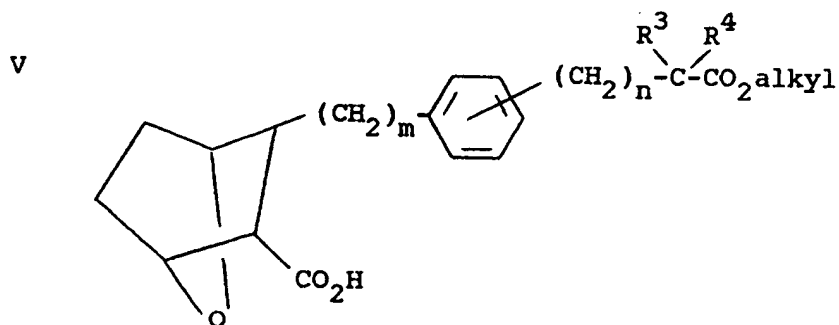
"Reagents for Organic Synthesis," Vol. 1, p. 142 (1967)) to form crude acid which is deacetylated and esterified by treatment with acidic alcohol such as methanolic HCl, to form the alcohol ester IV

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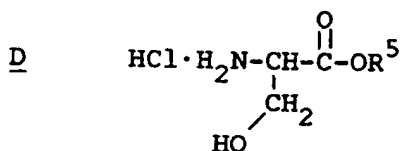
Next, the alcohol ester IV is subjected to a Jones oxidation to form the acid V

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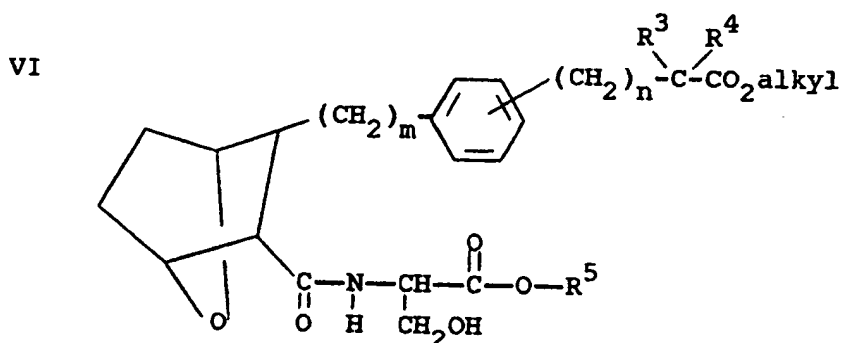


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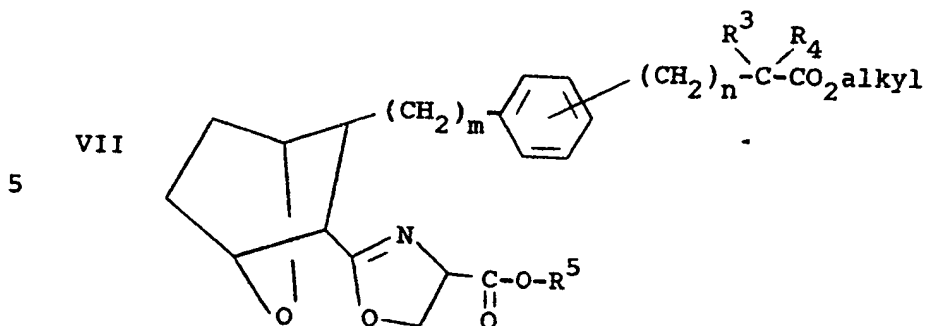
Acid V, in an inert organic solvent, such as tetrahydrofuran or dimethylformamide, is then made to undergo a carbodiimide coupling reaction with amine hydrochloride D



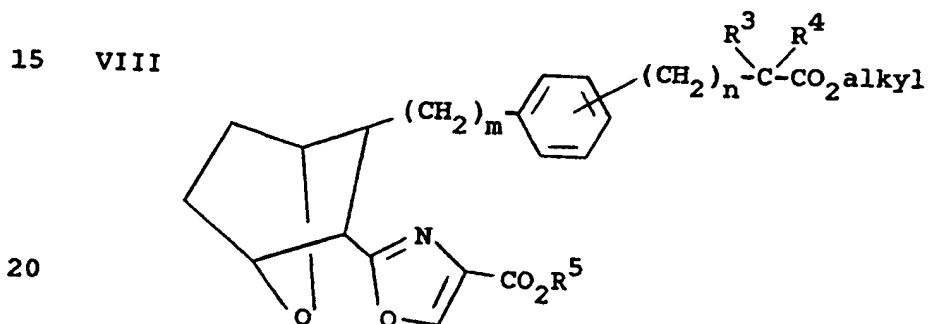
where R^5 is benzyl, in the presence of dicyclohexylcarbodiimide (DCC) or 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (WSC) and 1-hydroxybenzotriazole, and triethylamine, under an inert atmosphere such as argon employing a molar ratio of D:V of within the range of from about 1.2:1 to about 1:1, to form amide VI



Amide VI is then subjected to cyclodehydration wherein a solution of VI in an inert organic solvent such as tetrahydrofuran, acetonitrile or chloroform, under an inert atmosphere such as argon, is treated with triphenylphosphine (employing a molar ratio of VI:triphenylphosphine of from about 0.5:1 to about 1:1) and carbon tetrachloride in the presence of an amine base such as triethylamine or diisopropylethylamine, to form oxazoline VII



10 Oxazoline VII is oxidized by treatment with cupric bromide and 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) to form the oxazole VIII



The cupric bromide oxidation is carried out at a temperature of within the range of from about 20°C to about 50°C, employing a molar ratio of cupric bromide to VII of within the range of from about 2:1 to about 6:1 and a molar ratio of cupric bromide to DBU of within the range of from about 1:1 to about 1:3 in an inert organic solvent such as ethyl acetate, methylene chloride or preferably ethyl-acetate/chloroform (1:1, v/v).

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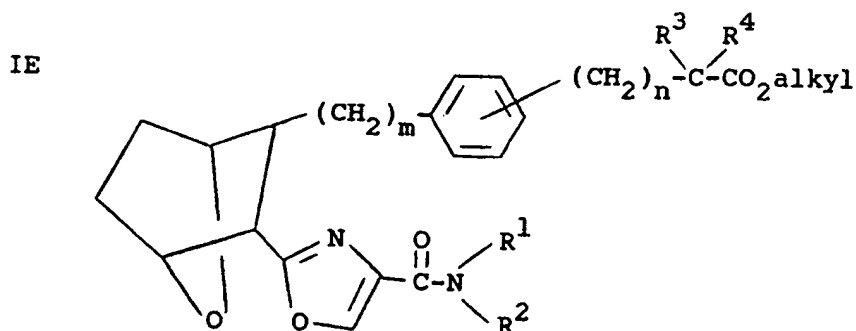
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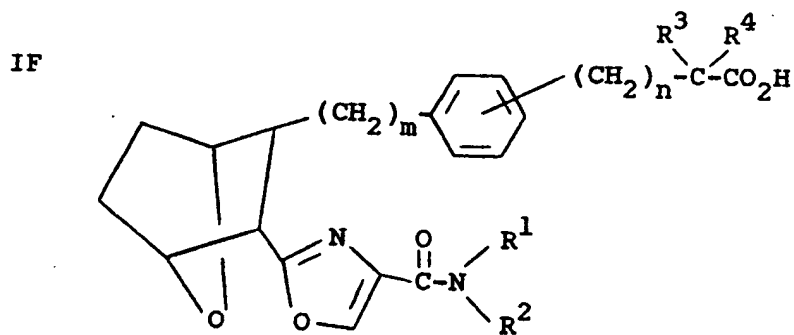
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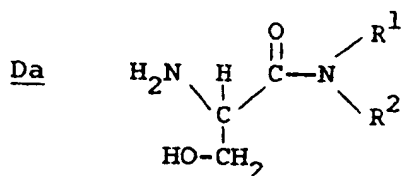
employing a molar ratio of E:IX of within the range of from about 1.1:1 to about 1.5:1, to form the oxazole IE



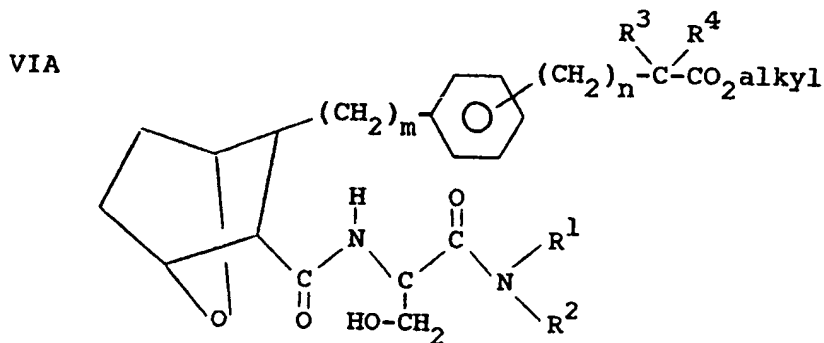
10 Oxazole X is hydrolyzed to the corresponding
acid XI by treating X with a base such as lithium
hydroxide, sodium hydroxide or potassium hydroxide
to form the corresponding alkali metal salt, followed
15 by neutralization with an acid such as dilute hydro-
chloric acid or oxalic acid to form acid IF.



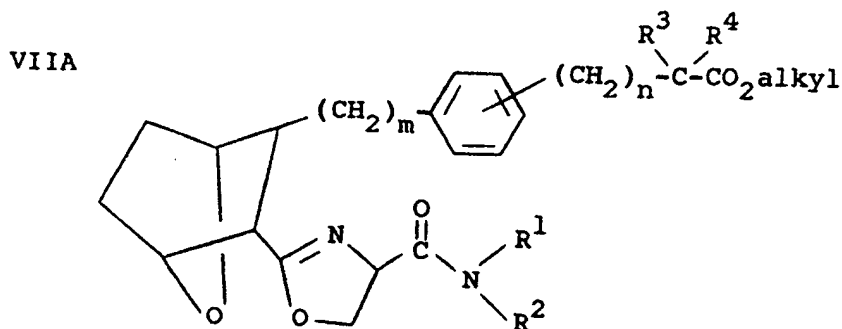
In an alternate preferred procedure for the
preparation of IF, acid V is made to undergo a
carbodiimide coupling reaction with amine Da



in the presence of dicyclohexylcarbodiimide or
 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide
 hydrochloride and 1-hydroxybenzotriazole and
 triethylamine as described hereinbefore to form
 hydroxy amide VIA



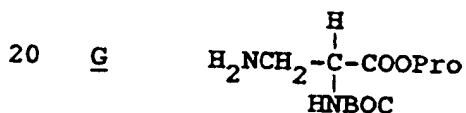
Hydroxy amide VIA is then subjected to
 cyclodehydration as described hereinbefore (with
 respect to the preparation of VII). A preferred
 method for this conversion involves treatment of
 VIA with an alkylsulfonyl chloride, such as
 methanesulfonyl chloride in the presence of an
 amine such as triethylamine or pyridine followed
 by treatment of the resulting alkylsulfonate
 intermediate with triethylamine in methylene
 chloride to form oxazoline VIIA



10 which is made to undergo oxidation as described
hereinbefore (with respect to the preparation of
VIII) to form oxazole IE.

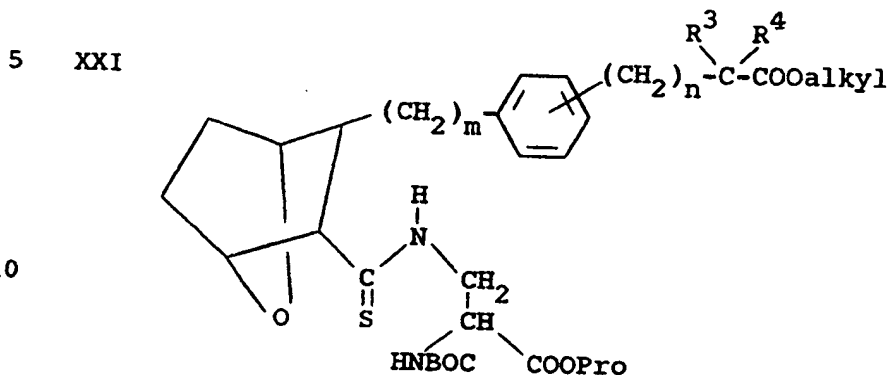
Ester IF may then be hydrolyzed by treatment
with an aqueous solution of alkali metal base and
15 then aqueous acid to form the corresponding acid IF.

Compounds of the invention where X is NH are
prepared starting with acid V which is made to
undergo a coupling reaction with amine G

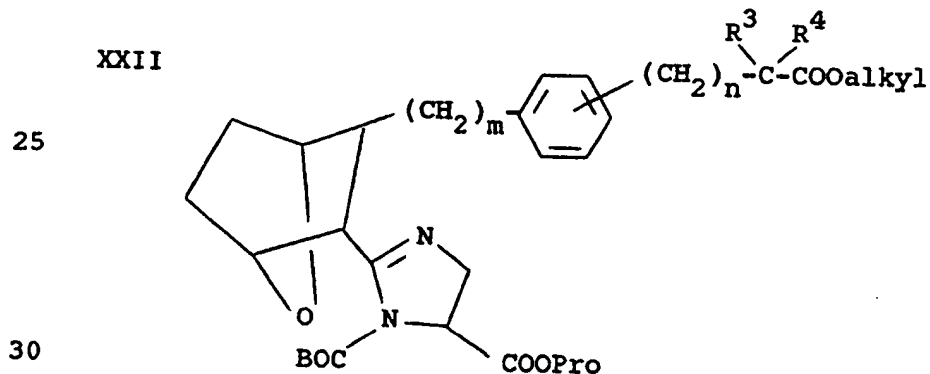


where BOC is t-butyloxycarbonyl and Pro is a
protecting group such as benzyl, in the presence of
a coupling agent such as 1-(3-dimethylaminopropyl)-
25 3-ethylcarbodiimide hydrochloride (WSC) and
1-hydroxybenzotriazole (HOBT) and methylene chloride
employing a molar ratio of V:G of within the range
of from about 1.2:1 to about 1:1, for a period of
from about 12 to about 90 hours. The resulting amide
30 is made to undergo a thionation reaction by treating
the amide with Lawesson's reagent in the presence of

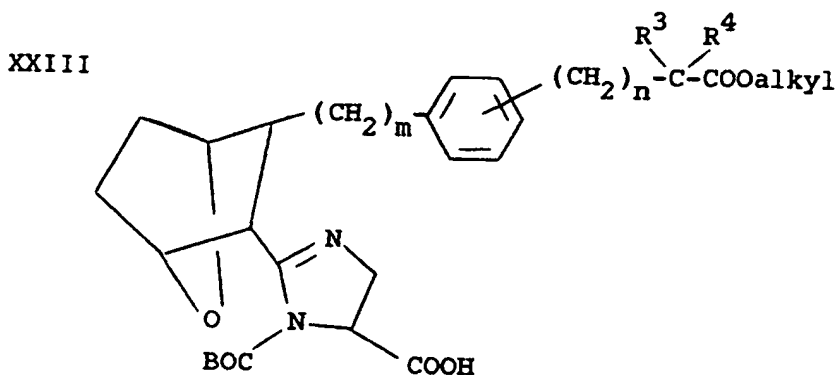
benzene at a temperature of from about 50 to about 75°C for a period of from about 1 to about 4 hours, to form the ester XXI



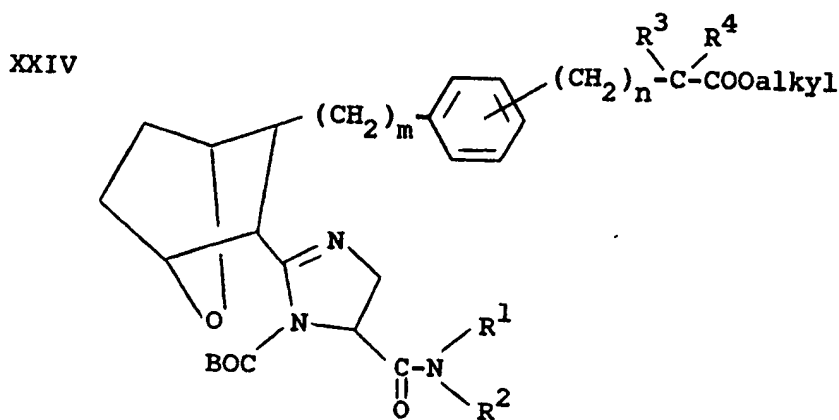
The ester XXI is cyclized by treating a solution of XXI in an inert organic solvent such as acetonitrile, chloroform or tetrahydrofuran with triphenylphosphine (employing a molar ratio of XXI:triphenylphosphine of from about 0.5:1 to about 1:1) and carbon tetrachloride in the presence of an amine base such as triethylamine or diisopropylethylamine, to form imidazoline XXII



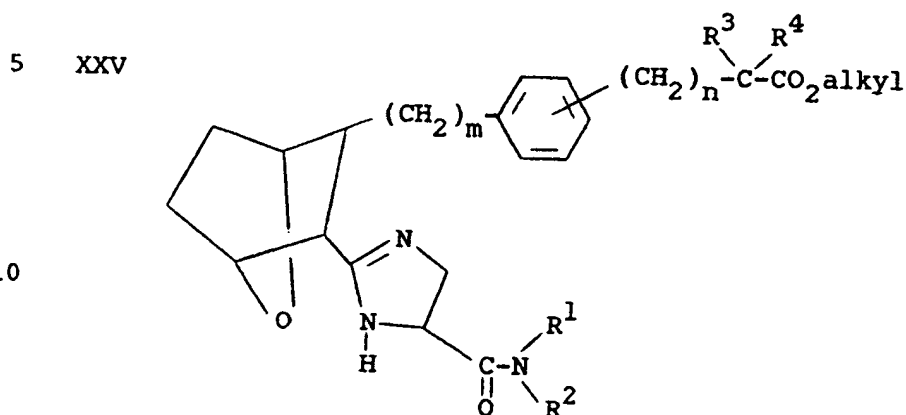
Imidazoline XXII is then deprotected to remove the Pro protecting group, using conventional procedures for example, by hydrogenation when Pro is benzyl, to form the acid XXIII



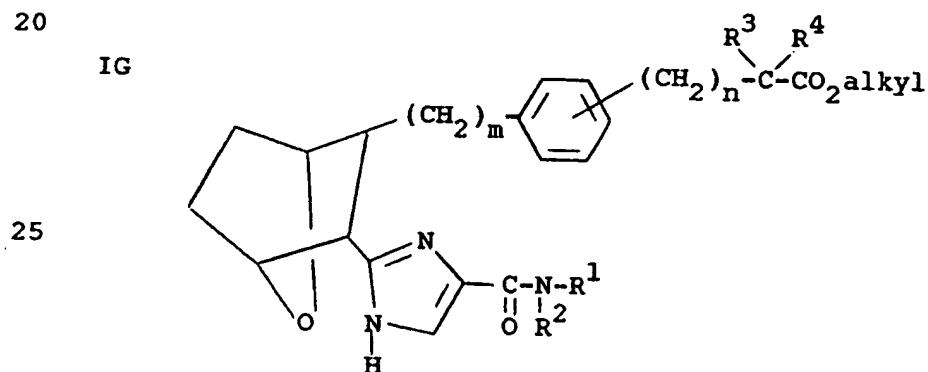
Next, the acid XXIII is made to undergo a coupling reaction with amine E in the presence of an amine base such as pyridine or triethylamine under an inert atmosphere such as argon in the presence of a coupling agent such as WSC and HOBT and chloroform, employing a molar ratio of E:XXIII of within the range of from about 0.8:1 to about 1.2:1 to form amide XXIV



The amine XXIV in solution in methylene chloride is then treated with trifluoroacetic acid to remove the BOC group and form amide XXV

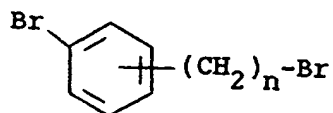


15 Amide XXV is oxidized by treatment with an oxidizing agent such as manganese dioxide in the presence of an inert organic solvent such as chloroform to form ester IG



The starting bromophenylalkyl alcohol A where n is 1, 2, 3 or 4 may be prepared by alkylating bromide K

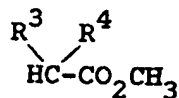
5 K



with an ester of the structure J

10

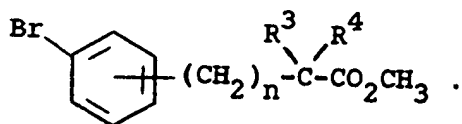
J



15 in the presence of lithium diisopropylamine and hexamethylphosphoric amide and an inert organic solvent such as tetrahydrofuran (THF) at reduced temperatures ranging from about -80°C up to about 0°C , employing a molar ratio of J:K of within the range of from about 1.2:1 to about 1:1, to form ester L

20

L

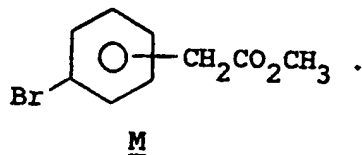


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Ester L where $n=0$ is prepared via alkylation of aryl ester M using standard methodology familiar

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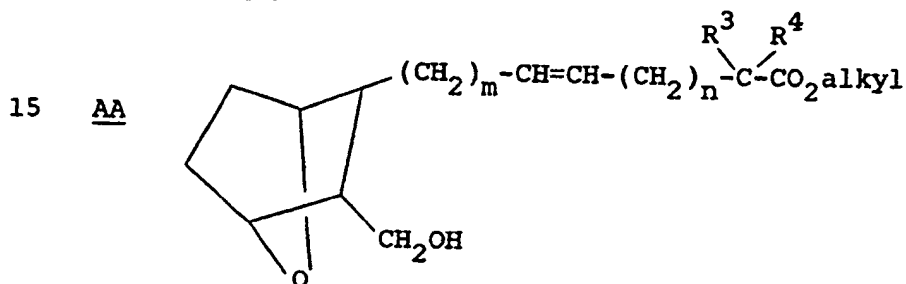
L



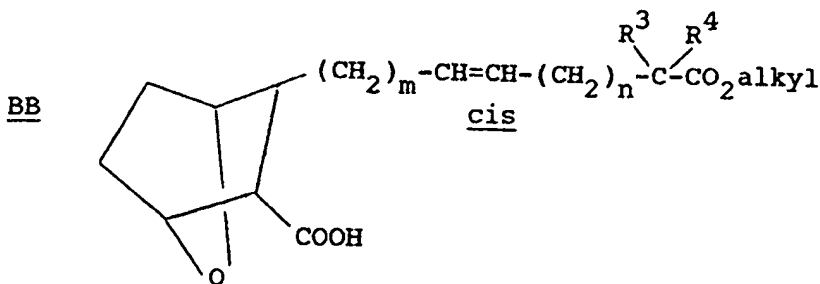
to those skilled in the art. Ester L is then hydrolyzed, for example, by treatment with aqueous alkali metal hydroxide and then reduced, for example, by treatment with borane-dimethylsulfide, to form the bromophenylalkyl alcohol A.

The compounds of formula I of the invention wherein Z is $-\text{CH}=\text{CH}-$ or $-(\text{CH}_2)_2-$ may be prepared as follows.

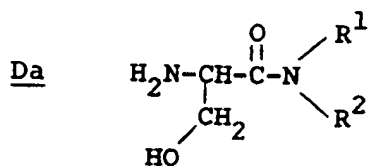
Compounds of the invention where Z is $-\text{CH}=\text{CH}-$ and preferably in the cis form, and X is O are prepared starting with the hydroxymethyl compound AA



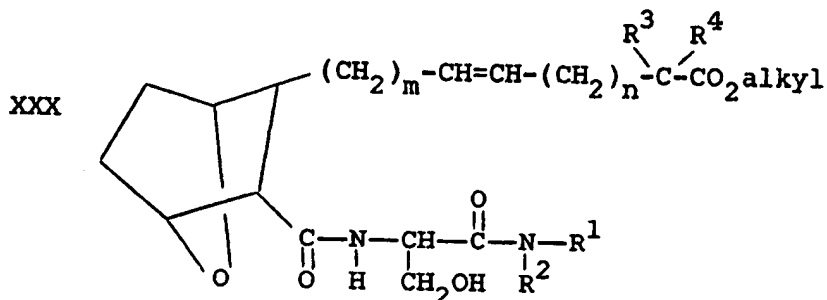
(which is prepared as described in U.S. Patent No. 4,143,054) which is subjected to a Jones oxidation wherein AA is reacted with Jones' Reagent (CrO_3 dissolved or suspended in aqueous sulfuric acid), in the presence of acetone, under an inert atmosphere such as argon at a temperature within the range of from about -10 to about 20°C , to form the corresponding carboxylic acid BB



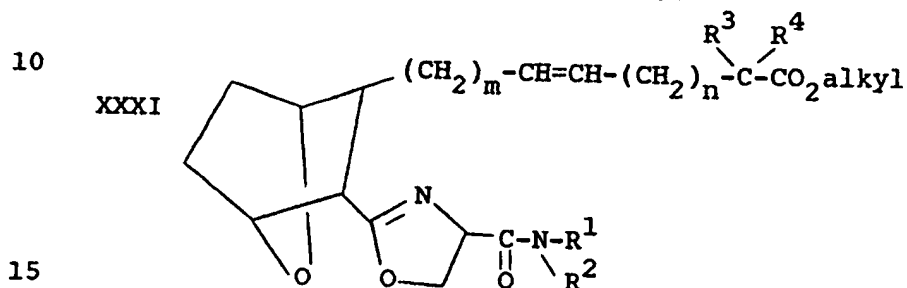
Acid BB, in an inert organic solvent, such as tetrahydrofuran, is then made to undergo a carbodiimide coupling reaction with amide Da



in the presence of dicyclohexylcarbodiimide (DCC) or 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (WSC) and 1-hydroxybenzotriazole under an inert atmosphere such as argon employing a molar ratio of Da:BB of within the range of from about 1.2:1 to about 1:1, to form hydroxybisamide XXX

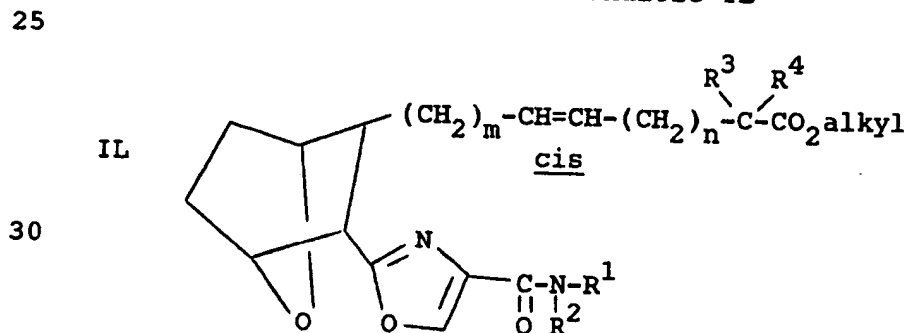


Hydroxybisamide XXX is then subjected to cyclodehydration wherein a solution of XXX in an inert organic solvent such as tetrahydrofuran, acetonitrile or chloroform, under an inert atmosphere such as argon, is treated with triphenylphosphine and carbon tetrachloride in the presence of an amine base such as triethylamine or diisopropylethylamine, to form oxazoline XXXI.

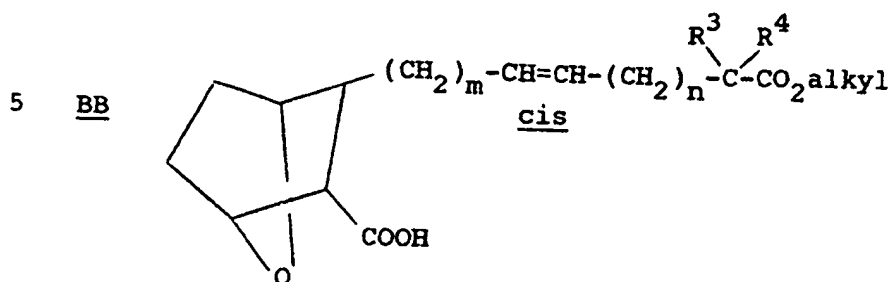


Alternatively, hydroxybisamide XXX is treated with a sulfonyl chloride, such as methane sulfonyl chloride, and an amine base such as triethylamine followed by treatment with potassium carbonate in acetone to form oxazoline XXXI.

Oxazoline XXXI is oxidized by treatment with manganese dioxide or nickel peroxide, preferably nickel peroxide, to form the oxazole IL



Alternatively, oxazole IDa can be prepared from acid BB

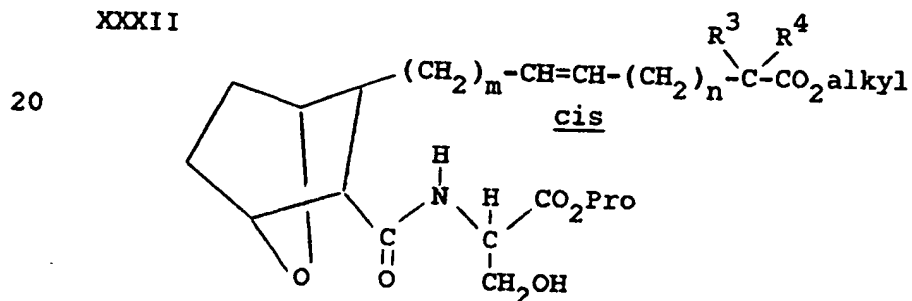


10

by a carbodiimide coupling as described previously except substituting CC for Da to obtain XXXII.



XXXII

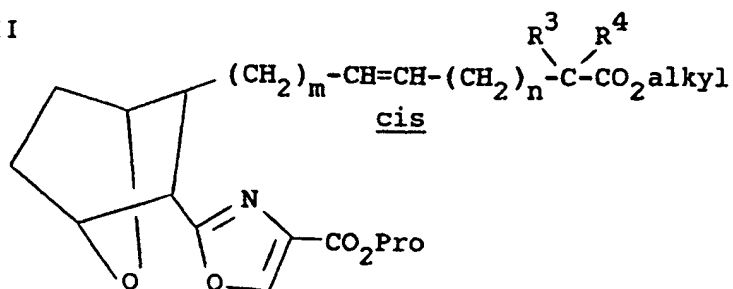


25

where Pro is a conventional protecting group.
Hydroxyamide XXXII is then subjected to a cyclodehydration and oxidation, as described for XXX and XXXI, to form XXXIII

XXXIII

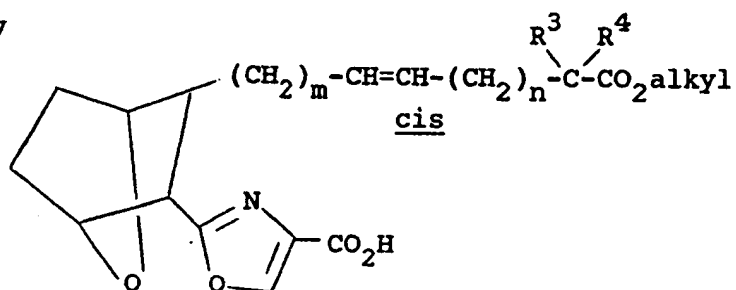
5



10 The protecting group of XXXIII can be removed to form the corresponding acid XXXIV

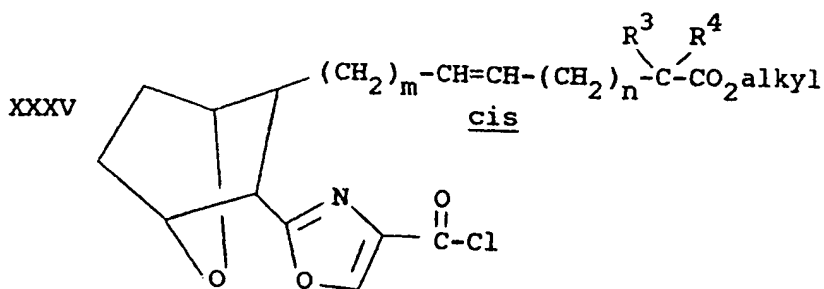
XXXIV

15

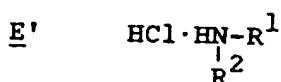


20 which is treated with excess oxalyl chloride in the presence of an inert organic solvent such as toluene, methylene chloride, or chloroform, and optionally a catalytic amount of dimethylformamide, while stirring under an inert atmosphere such as argon, to form the

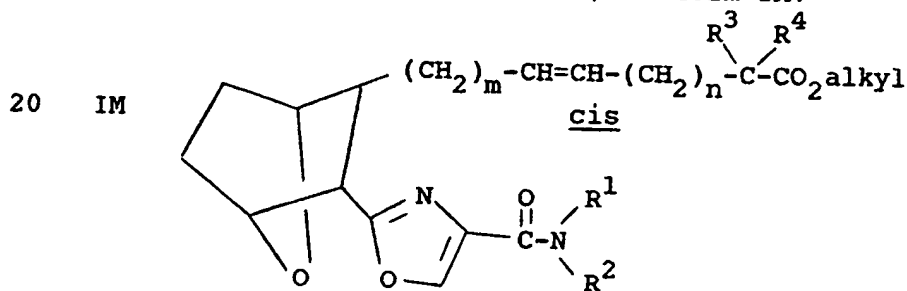
25 crude acid chloride XXXV



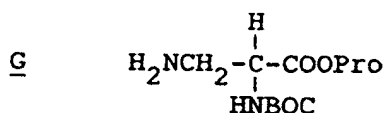
which is treated with amine hydrochloride E'



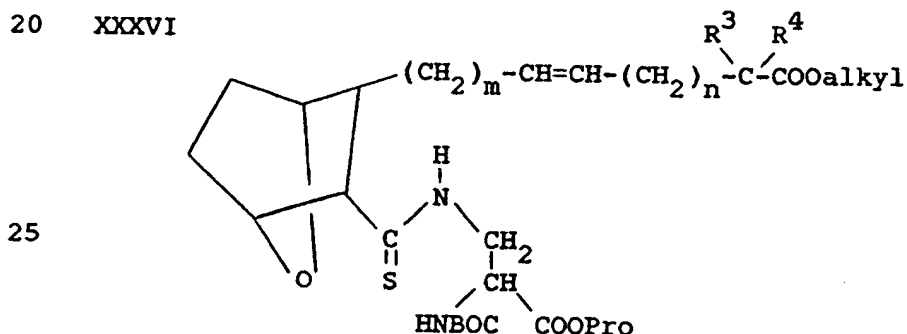
in the presence of an organic base such as triethyl-
 amine under an inert atmosphere such as argon,
 employing a molar ratio of XXXV:E' of within the
 range of from about 0.5:1 to about 1:1 and preferably
 from about 0.8:1 to about 1:1, to form IM.



Compounds of the invention IF where X is NH
 and Z¹ is -CH=CH- are prepared starting with acid
BB which is made to undergo a coupling reaction with
 amine G



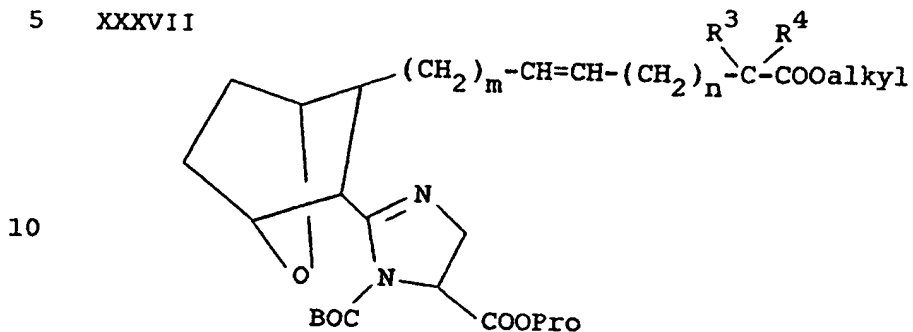
5 where BOC is t-butyloxycarbonyl and Pro is a protecting group, preferably benzyl, in the presence of a coupling agent such as 1-(3-dimethylamino-propyl)-3-ethylcarbodiimide hydrochloride (WSC) and
10 1-hydroxybenzotriazole (HOBT) and methylene chloride employing a molar ratio of BB:Q of within the range of from about 1.2:1 to about 1:1, for a period of from about 12 to about 90 hours. The resulting amide is made to undergo a thionation reaction by
15 treating the amide with Lawesson's reagent in the presence of benzene at a temperature of from about 50 to about 75°C for a period of from about 1 to about 4 hours, to form the ester XXXVI



30 The ester XXXVI is cyclized by treating a solution of XXXVI in an inert solvent such as acetonitrile, chloroform or tetrahydrofuran, with triphenylphosphine (employing a molar ratio of XXXVI:triphenylphosphine of from about 0.8:1 to about 0.5:1) and

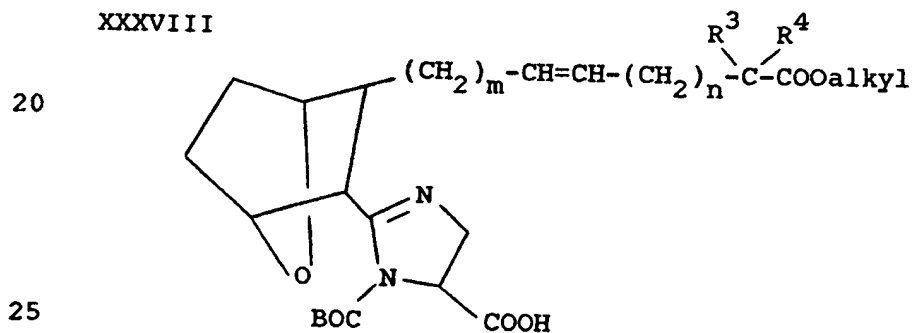
carbon tetrachloride in the presence of an amine base such as triethylamine or diisopropylethylamine, to form imidazoline XXXVII

5 XXXVII



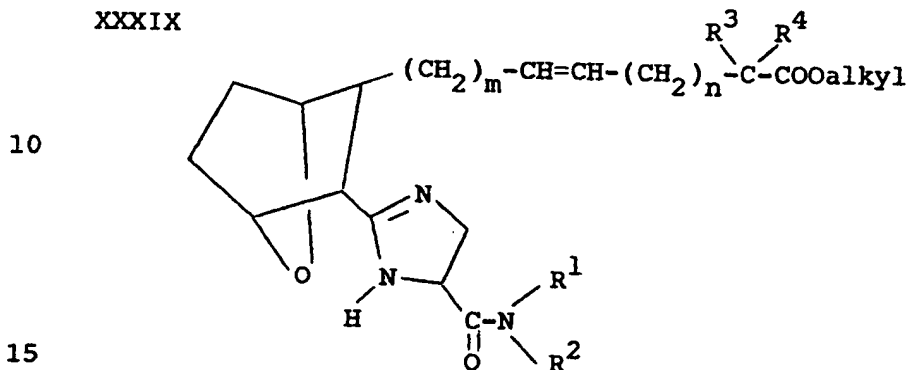
15 Imidazoline XXXVII is then deprotected to remove the Pro protecting group, using conventional procedures, to form the acid XXXVIII

20 XXXVIII

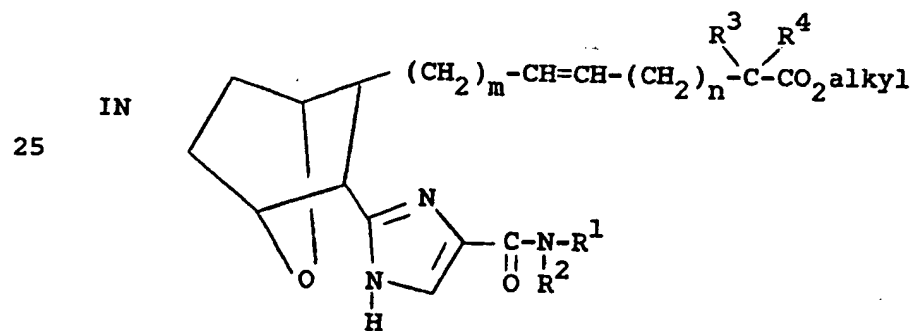


30 Next, the acid XXXVIII is made to undergo a coupling reaction with amine E in the presence of an amine base such as pyridine or triethylamine under an inert atmosphere such as argon in the presence

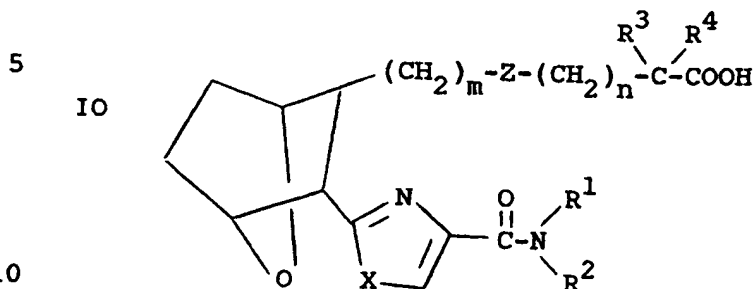
of a coupling agent such as WSC and HOBT and chloro-
form, employing a molar ratio of E:XXXVIII of within
the range of from about 0.8:1 to about 1.2:1 to form
amide XXXIX after removal of the BOC protecting group
5 with trifluoroacetic acid



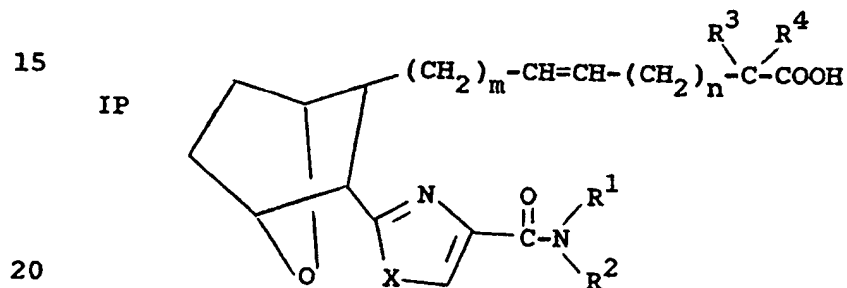
Amide XXXIX is oxidized by treatment with an
oxidizing agent such as manganese dioxide in the
presence of an inert solvent such as chloroform to
20 form ester IN



The aforementioned esters of the invention may be converted to the corresponding acids, that is IO



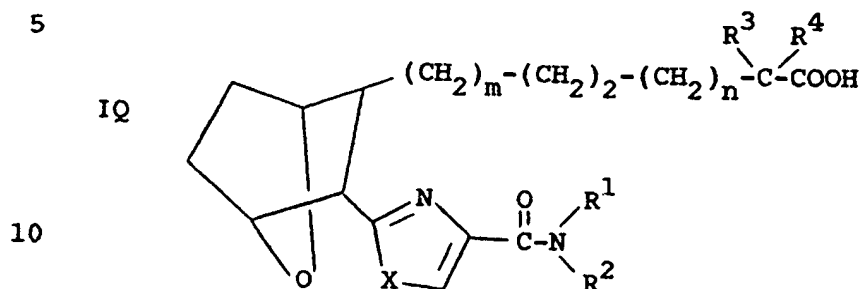
or



by treating the esters with a base, such as lithium hydroxide, sodium hydroxide or potassium hydroxide to form the corresponding alkali metal salt, followed by neutralization with an acid, such as dilute hydrochloric acid or oxalic acid to form the acid compounds of the invention.

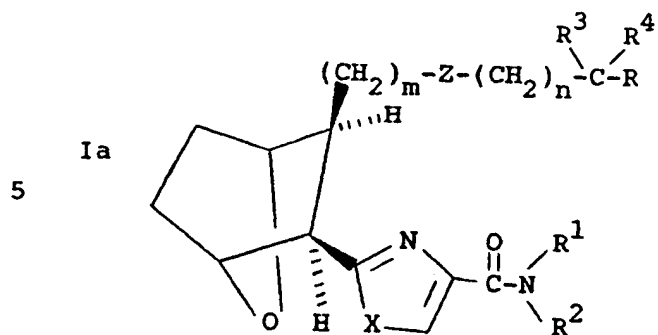
Compounds of formula I wherein Z is $-(CH_2)_2-$ may be prepared from acid IP by subjecting IP to hydrogenation using, for example, a hydrogenation

catalyst, such as palladium on carbon, in an inert organic solvent such as ethyl acetate (EtOAc) or acetic acid (AcOH) to form acid of the invention IQ

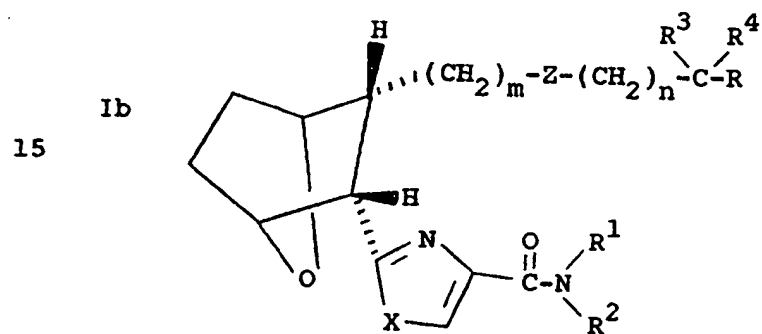


The compounds of this invention have four centers of asymmetry as indicated by the asterisks in formula I. However, it will be apparent that each of the formulae set out above which do not include asterisks still represent all of the possible stereoisomers thereof. All of the various stereoisomeric forms are within the scope of the invention.

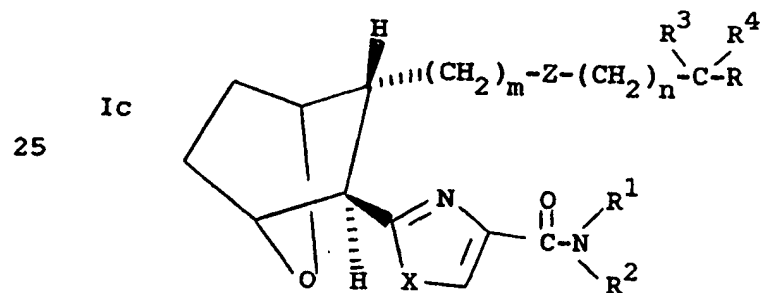
The various stereoisomeric forms of the compounds of the invention, namely, cis-exo, cis-endo and all trans forms and stereoisomeric pairs may be prepared by employing starting materials and following the procedures as outlined in U.S. Patent No. 4,143,054. Examples of such stereoisomers are set out below.



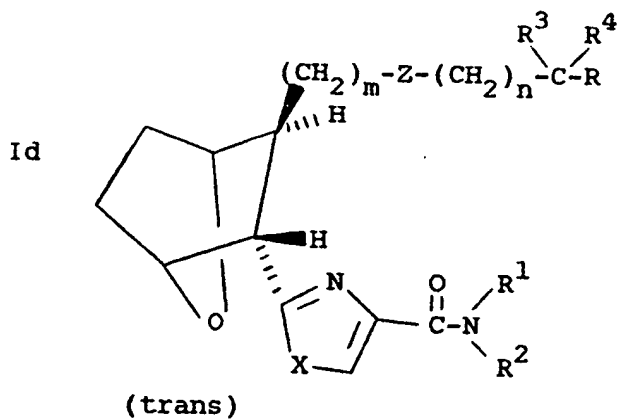
10 (cis-endo)



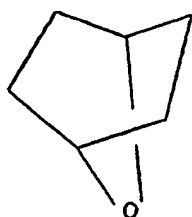
20 (cis-exo)



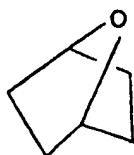
30 (trans)



The nucleus in each of the compounds of the invention is depicted as



for matter of convenience; it will also be appreciated that the nucleus in the compounds of the invention may be depicted as



The compounds of this invention are thromboxane receptor antagonists and as such are useful as inhibitors of thromboxane receptor mediated actions. The term "thromboxane receptor

antagonist" includes compounds which are so-called thromboxane A₂ receptor antagonists, thromboxane A₂ antagonists, thromboxane A₂/prostaglandin endoperoxide antagonists, TP-receptor antagonists, or thromboxane antagonists.

The compounds of the invention may also be thromboxane synthetase inhibitors and thus may be useful as inhibitors of thromboxane production.

The compounds of this invention are useful as inhibitors of platelet function, i.e., for the prevention and treatment of thrombotic vascular occlusive disorders, whether complete or partial, for example, arterial thrombosis, including that of the coronary, cerebral, ophthalmic, hepatic, mesenteric, renal, peripheral arteries or vascular or organ grafts, unstable angina, transient ischemic attacks, or intermittent claudication. They may be useful to prevent thrombosis following vascular injury produced in the course of diagnostic or therapeutic procedures such as endarterectomy or angiography. The compounds may be useful in the treatment or prevention of disorders characterized by platelet consumption and/or activation, including, platelet activation, dysfunction, and/or loss during extracorporeal circulation, the use of radiographic contrast agents, thrombotic thrombocytopenia purpura, disseminated intravascular coagulation, purpura fulminans, hemolytic transfusion reaction, or hemolytic uremic syndrome, systemic lupus, cyclosporine-induced renal toxicity, pulmonary hypertension, side effects from dialysis, or abdominal aortic aneurism repair. The compounds may be used in the treatment of venous thrombosis

or embolism, including pulmonary embolism, deep venous thrombosis, hepatic vein thrombosis, and renal vein thrombosis.

5 The compounds of this invention are useful
as inhibitors of arterial or venous vasoconstriction.
Accordingly, they may be useful to prevent vasoconstriction associated with unstable angina, chronic stable angina, and variant, or Prinzmetal's angina, Raynaud's syndrome, migraine headache, vasospasm of
10 the coronary, cerebral, ophthalmic, hepatic, mesenteric, renal, peripheral arteries or vascular grafts, vascular injury such as that associated with surgery or trauma. Hypertension of pregnancy, the hepato-renal syndrome, and pulmonary hypertension are additional examples of vasoconstrictive
15 disorders treatable by the compounds of this invention.

 The compounds of this invention are useful as inhibitors of bronchoconstriction, i.e., airway
20 hyperresponsiveness, allergic bronchospasm, asthma, and bronchoconstrictive responses to environmental, infectious, noxious or mechanical stimuli.

 The compounds of this invention are useful
25 as inhibitors of ischemic and reperfusion injury to various tissues, including, myocardium, skin, brain, bowel, or kidney, alone or in combination with other agents intended to restore blood flow. For example, these compounds may be useful for
30 improving postischemic myocardial function and decreasing myocardial infarct size. Ischemia caused by reduced blood flow during diagnostic or therapeutic procedures may benefit by treatment

with these compounds, for example, they reduce the myocardial stunning observed after bypass surgery. In addition, they may be useful for reducing the tissue injury caused by a stroke.

5 The compounds of this invention may be useful in the prevention or treatment of other conditions including burns, diabetic retinopathy, tumor metastases and tardive dyskinesia. The compounds may be useful in potentiating diuretic-
10 induced diuresis.

 In addition, the thromboxane receptor antagonists of the invention may be used with a thrombolytic agent such as t-PA, streptokinase, urokinase, prourokinase or anisoylated plasminogen-
15 streptokinase activator complex (APSAC) within 6 hours of a myocardial infarction. In such case, the thrombolytic agent may be used in amounts conventionally employed, for example, as disclosed in the Physicians' Desk Reference for reducing
20 post-ischemic myocardial injury.

 The compounds of the invention can be administered orally or parenterally to various mammalian species known to be subject to such
maladies, e.g., humans, cats, dogs and the like in
25 an effective amount within the dosage range of about 0.1 to about 100 mg/kg, preferably about 0.2 to about 50 mg/kg and more preferably about 0.5 to about 25 mg/kg (or from about 1 to about 2500 mg, preferably from about 5 to about 2000 mg) on a
30 regimen in single or 2 to 4 divided daily doses.

 The active substance can be utilized in a composition such as tablet, capsule, solution or suspension containing about 5 to about 500 mg per

unit of dosage of a compound or mixture of compounds of formula I or in topical form for wound healing (0.01 to 5% by weight compound of formula I, 1 to 5 treatments per day). They may
5 be compounded in conventional matter with a physiologically acceptable vehicle or carrier, excipient, binder, preservative, stabilizer, flavor, etc., or with a topical carrier such as Plastibase (mineral oil gelled with polyethylene)
10 as called for by accepted pharmaceutical practice. Also as indicated in the discussion above, certain members additionally serve as intermediates for other members of the group.

It will be appreciated that the acid or
15 ester compounds of the invention may be converted to any pharmaceutically acceptable salt, employing conventional procedures, to facilitate preparation of pharmaceutical compositions containing such compounds.

20 The compounds of the invention may also be administered topically to treat peripheral vascular diseases and as such may be formulated as a cream or ointment.

25 The following Examples represent preferred embodiments of the present invention. Unless otherwise indicated, all temperatures are expressed in degrees Centigrade.

Example 1

[1S-(1 α , 2 α , 3 α , 4 α)]-2-[[3-[4-[(Pentylamino)carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl]methyl]- α , α -dimethylbenzenepropanoic acid, methyl ester

5

A. 2-Bromo- α , α -dimethylbenzenepropanoic acid, methyl ester

To a solution of 17 mL (120 mmol, distilled from calcium hydride) of diisopropylamine in 100 mL of dry THF (distilled from sodium/benzophenone) cooled to -78° was added dropwise 44 mL (2.5M in hexanes, 110 mmol, Aldrich) of n-butyllithium solution over ~15 minutes. The reaction mixture was stirred for an additional 30 minutes then a solution of 10.2 g (100 mmol, Aldrich) of methyl isobutyrate in 10 mL of THF was added dropwise over 15 minutes. After 30 minutes 17 mL (98 mmol, distilled from calcium hydride) of hexamethyl phosphoric triamide (HMPA) was added followed by a solution of 25.0 g (100 mmol, Aldrich) of 2-bromobenzyl bromide in 10 mL of THF over ~5 minutes. The reaction mixture was stirred at -78° for 1 hour (h) then stored at 0° for 18 h. The resulting solution was quenched by addition of 5 mL of water then concentrated in vacuo. The residue was partitioned between 150 mL of 1M aqueous HCl solution and 150 mL of diethyl ether(ether). The organic layer was separated, washed with two-150 mL portions of water, 50 mL of brine, dried (magnesium sulfate) and concentrated in vacuo to give a yellow oil. The crude material was purified by flash chromatography (Merck silica, 120x10 cm, 1:19 ether/hexane) to afford 22.2 g (81.9 mmol, 82%) of title ester as a pale yellow liquid.

B. 2-Bromo- α,α -dimethylbenzenepropanol

A solution of 21.3 g (78.6 mmol) of Part A ester in 50 mL of THF (distilled from sodium/benzophenone) and 50 mL of 3M aqueous NaOH solution was stirred at 65° for 18 h then cooled and concentrated in vacuo. The residue was cooled in an ice-bath and acidified by slow addition of 15 mL of conc. HCl. The resulting slurry was partitioned between 100 mL of water and 50 mL of ether. The aqueous layer was separated and extracted with an additional 50 mL of ether. The ether extracts were combined, washed with 50 mL of brine, dried (magnesium sulfate) and concentrated in vacuo to give 19.8 g of the crude acid as a white solid, mp 84-86°.

To a solution of the crude acid in 100 mL of dry THF (distilled from sodium/benzophenone) cooled in an ice-bath was added dropwise 44 mL (2.0M in THF, 88 mmol, Aldrich) of borane-dimethylsulfide solution over 20 minutes. The reaction mixture was stirred at 0° for 2 h then at room temperature for 20 h. The resulting solution was quenched by addition of 5 mL of water, stirred for 30 minutes, then concentrated in vacuo. The residue was partitioned between 100 mL of 1M aqueous HCl solution and 100 mL of ether. The organic layer was separated, washed with two-100 mL portions of 1M aqueous NaOH solution, 50 mL of brine, dried (magnesium sulfate) and concentrated in vacuo to give an oil. The oil was dissolved in 50 mL of anhydrous methanol and concentrated in vacuo; repeated with an additional 50 mL of methanol to

afford 18.6 g (76.5 mmol, 97% from Part A ester) of title alcohol as a colorless oil.

5 C. Bromo[[dimethyl(1,1,2-trimethylpropyl)-
silyl]oxy]benzene

To a solution of 18.6 g (76.5 mmol) of Part B alcohol, 15.0 g (84.3 mmol, Aldrich) of dimethylthexylsilyl chloride and 14 mL (100 mmol, distilled from calcium hydride) of triethylamine in 100 mL of methylene chloride (distilled from phosphorous pentoxide) was added 1.84 g (15.1 mmol, Aldrich) of 4-dimethylaminopyridine at room temperature. The reaction mixture was stirred for 20 h then cooled in an ice-bath and diluted with 100 mL of hexane to precipitate triethylamine hydrochloride. After 15 minutes the slurry was filtered. The filtrate was concentrated in vacuo and the residue partitioned between 10 mL of hexane and 100 mL of 1M aqueous HCl solution. The organic layer was separated, washed with an additional 100 mL of 1M aqueous HCl solution, 100 mL of water, dried (magnesium sulfate) and concentrated in vacuo to give an oil. The crude material was purified by flash chromatography (Merck silica, 12x10 cm, 1:99 ether/hexane) to afford 23.9 g (62.1 mmol, 81%) of title silyl ether as a colorless liquid.

30 D. [1S-(1 α ,2 α (R*),3 α ,4 α)]- α -[2-[3-[[Dimethyl(1,1,2-trimethylpropyl)silyl]oxy]-2,2-dimethylpropyl]phenyl]-7-oxabicyclo[2.2.1]-
heptane-2,3-dimethanol

To 1.00 g (41 mmol, Mallinckrodt) of hammer-crushed magnesium turnings covered with 10 mL of dry

THF (distilled from sodium/benzophenone) was added a small crystal of iodine, 150 μ L of 1,2-dibromoethane and then $\sim 1/2$ of 12.0 g (31.2 mmol) of title aryl bromide was added in one portion. The mixture was warmed until a reaction started then the remainder of the aryl bromide was added dropwise rapidly followed by heating to reflux (120° oil bath) for 2 h. The resulting Grignard solution was cooled to room temperature then 20 mL of THF was added to solubilize precipitated reagent.

To a solution of 4.06 g (26.0 mmol) of [3aR-(3 α ,4 β ,7 β ,7 α)]-octahydro-4,7-epoxyisobenzofuran-1-ol in 20 mL of dry THF cooled in an ice-bath was added dropwise 12.5 mL (2.0M in THF, 25 mmol, Aldrich) of ethylmagnesium bromide over 15 minutes. The reaction mixture was stirred for 15 minutes then the aryl Grignard solution (~ 31 mmol) from above was added via cannula over ~ 10 minutes. The resulting solution was warmed to room temperature, stirred for 18 h then cooled to 0° and quenched by careful addition of 5 mL of water followed by slow addition of 200 mL of 10% aqueous ammonium chloride solution. The mixture was extracted with two-100 mL portions of ethyl acetate. The organic extracts were combined, washed with 50 mL of brine, dried (magnesium sulfate) and concentrated in vacuo to give an oil. The crude material was purified by flash chromatography (Merck silica, 20x5.0 cm, 1:4 ethyl acetate/hexane then 4:1 ethyl acetate/hexane) to afford 9.60 g (20.8 mmol, 83%) of title diol as a colorless oil.

E. [1S-(1 α ,2 α ,3 α ,4 α)]-2-[[2-[3-[[Dimethyl-(1,1,2-trimethylpropyl)silyl]oxy]-2,2-dimethylpropyl]phenyl]methyl]-7-oxabicyclo-[2.2.1]heptane-3-methanol

5 A mixture of 9.50 g (20.6 mmol) of Part D diol and 3.2 g of palladium hydroxide on carbon catalyst (<50% water, Aldrich) in 75 mL of glacial acetic acid was stirred under an atmosphere of hydrogen (balloon) for 18 h. The reaction was
10 filtered to remove the catalyst. The filtrate was concentrated by rotoevaporation (room temperature bath/oil pump vacuum) to give an oil. The oil was dissolved in 50 mL of toluene and concentrated in vacuo to remove residual acetic acid; repeated with
15 an additional 50 mL of toluene. The crude material was purified by flash chromatography (Merck silica, 20x5.0 cm, 2:3 ethyl acetate/hexane) to afford 8.23 g (18.5 mmol, 90%) of title alcohol as a colorless oil.

20

F. [1S-(1 α ,2 α ,3 α ,4 α)]-2-[[2-[3-[[Dimethyl-(1,1,2-trimethylpropyl)silyl]oxy]-2,2-dimethylpropyl]phenyl]methyl]-7-oxabicyclo-[2.2.1]heptane-3-methanol, acetate

25 To a solution of 8.20 g (18.4 mmol) of Part E alcohol in 20 mL of pyridine (Burdick and Jackson) and 20 mL of reagent acetic anhydride (Mallinckrodt) was added at room temperature 112 mg (0.92 mmol, Aldrich) of 4-dimethylaminopyridine. The reaction
30 mixture was stirred for 1 h then concentrated in vacuo and the residue partitioned between 50 mL of ethyl acetate and 50 mL of ice-cold 1M aqueous HCl solution. The organic layer was separated, washed

with an additional 50 mL of 1M aqueous HCl solution, 50 mL of water, 25 mL of brine, dried (magnesium sulfate) and concentrated in vacuo to give 8.83 g (18.1 mmol, 98%) of title acetate as a colorless oil.

G. [1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-(Hydroxymethyl)-7-oxabicyclo[2.2.1]hept-2-yl]methyl-(α , α -dimethylbenzenepropanoic acid, methyl ester

10 To a solution of 8.80 g (18.0 mmol) of Part F acetate in 100 mL of reagent acetone in an ambient water bath was added rapidly 30 mL (2.6M in Cr⁺⁶, prepared as described in Fieser and Fieser, "Reagents for Organic Synthesis," Vol.1, p. 142) of Jones
15 reagent. The reaction mixture was stirred for 2 h then quenched by addition of 5 mL of isopropanol followed by stirring for 15 minutes. The resulting slurry was filtered through a pad of Celite to remove precipitated chromium salts. The filtrate
20 was concentrated in vacuo and the residue partitioned between 50 mL of ether and 100 mL of 1M aqueous HCl solution. The aqueous layer was separated and extracted with an additional 50 mL of ether. The organic layers were combined, washed
25 with 50 mL of brine, dried (magnesium sulfate) and concentrated in vacuo to give the crude acid-acetate as an oil. To the crude material was added 100 mL of an ice-cold solution of acidic methanol (prepared by addition of 2 mL of acetyl chloride to 100 mL of
30 anhydrous methanol at 0°) then stirred at room temperature for 64 h. The solution was concentrated in vacuo and the resulting oil purified by flash

chromatography (Merck silica, 20x5.0 cm, 3:2 ethyl acetate/hexane) to afford 4.31 g (13.1 mmol, 73%) of title alcohol-ester as a colorless oil.

5 H. [1S-(1 α ,2 α ,3 α ,4 α)]-2-[(3-Carboxy-7-oxa-bicyclo[2.2.1]hept-2-yl)methyl]- α , α -dimeth-
ylbenzenepropanoic acid, methyl ester

To a solution of 2.00 g (6.06 mmol) of Part
G alcohol-ester in 25 mL of reagent acetone in an
10 ambient water bath was added dropwise 6.0 mL (2.6M
in Cr⁺⁶) of Jones reagent. The reaction mixture
was stirred for 1.5 h then quenched by addition of
5 mL of isopropanol and stirred for 15 minutes.
The resulting slurry was filtered through a pad of
15 Celite to remove precipitated chromium salts. The
filtrate was concentrated in vacuo and the residue
partitioned between 50 mL of ether and 50 mL of 1M
aqueous HCl solution. The aqueous layer was
separated and extracted with an additional 50 mL
20 of ether. The organic layers were combined, washed
with 50 mL of brine, dried (magnesium sulfate) and
concentrated in vacuo to give an oil. The crude
material was purified by flash chromatography (Merck
silica, 15x5.0 cm, 1:19 methanol/methylene chloride)
25 to afford 1.92 g (5.58 mmol, 92%) of title acid as a
colorless glass.

I. [1S-[1 α ,2 α ,3 α (R*),4 α]]-2-[[3-[[[1-(Hydroxymethyl)-2-oxo-2-(phenylmethoxy)ethyl]-amino]carbonyl]-7-oxabicyclo[2.2.1]hept-2-yl]methyl]- α,α -dimethylbenzenepropanoic acid, methyl ester

To a solution of 1.78 g (5.17 mmol) of Part H acid in 15 mL of DMF (Burdick and Jackson) cooled in an ice-bath was added 960 mg (80%, 5.7 mmol, Aldrich) of 1-hydroxybenzotriazole hydrate, 1.32 g (5.69 mmol, Sigma) of L-serine benzyl ester hydrochloride, 0.80 mL (5.7 mmol, distilled from calcium hydride) of triethylamine then 1.09 g (5.69 mmol, JBL) of WSC. The reaction mixture was stirred at 0° for 2 h then at room temperature for 16 h. The resulting slurry was partitioned between 30 mL of ethyl acetate and 60 mL of 1M aqueous HCl solution. The aqueous layer was separated and extracted with an additional 30 mL of ethyl acetate. The organic extracts were combined, dried (magnesium sulfate) and concentrated in vacuo to give an oil. The crude material was purified by flash chromatography (Merck silica, 20x5.0 cm, ethyl acetate) to afford 2.50 g (4.80 mmol, 93%) of title amide as a colorless glass.

J. [1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-[4,5-Dihydro-4-[(phenylmethoxy)carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl]methyl]- α,α -dimethylbenzenepropanoic acid, methyl ester

A mixture of 2.45 g (4.70 mmol) of Part I amide and 1.85 g (7.06 mmol, Aldrich) of triphenylphosphine in 20 mL of acetonitrile (Burdick and Jackson) was stirred at room temperature until homogeneous then 1.25 mL (7.2 mmol, Aldrich) of diiso-

propylethylamine and 0.70 mL (7.2 mmol, Mallinckrodt) of reagent carbon tetrachloride were added. The reaction mixture was stirred for 2.5 h then partitioned between 75 mL of saturated sodium bicarbonate solution and 25 mL of ethyl acetate. The aqueous layer was separated and extracted with an additional 25 mL of ethyl acetate. The organic extracts were combined, dried (sodium sulfate) and concentrated in vacuo to give an oily solid. The crude material was purified by flash chromatography (Merck silica, 20x5.0 cm, 2:1 ethyl acetate/hexane) to afford 2.05 g (4.08 mmol, 87%) of title oxazoline as a solid.

K. [1S-(1 α ,2 α ,3 α ,4 α)]- α , α -Dimethyl-2-[[3-[4-[(phenylmethoxy)carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl)methyl]-benzenepropanoic acid, methyl ester

To a solution of 2.66 g (17.5 mmol, Aldrich) of DBU in 20 mL of ethyl acetate (Burdick and Jackson) was added 1.95 g (8.74 mmol, Aldrich) of copper(II) bromide. After 10 minutes a solution of 2.00 g (3.98 mmol) of Part J oxazoline in 20 mL of chloroform (Burdick and Jackson) was added over 5 minutes. The addition was mildly exothermic. The reaction mixture was stirred for 16 h then added was 100 mL of 1:1 saturated aqueous ammonium chloride/concentrated ammonium hydroxide solution and extracted with 50 mL of ether. The aqueous layer was separated and extracted with an additional 25 mL of ether. The combined ether layers were washed with 50 mL of brine, dried (magnesium sulfate) and concentrated in vacuo to give an oil. The crude

material was purified by flash chromatography (Merck silica, 20x5.0 cm, 1:1 ethyl acetate/hexane) to give 645 mg (1.29 mmol, 32%) of title oxazole as a white solid, mp 89-91°.

5

L. [1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-(4-Carboxy-2-oxazolyl)-7-oxabicyclo[2.2.1]hept-2-yl]-methyl]- α , α -dimethylbenzenepropanoic acid, methyl ester

10

A mixture of 640 mg (1.28 mmol) of Part K benzyl ester and 64 mg of 20% palladium hydroxide on carbon catalyst (<50% water, Aldrich) in 10 mL of ethyl acetate was stirred rapidly under an atmosphere of hydrogen (balloon) for 2 h. The reaction mixture was passed through a 0.4 μ M polycarbonate membrane to remove the catalyst. The filtrate was concentrated in vacuo to afford 490 mg (1.19 mmol, 93%) of title oxazole acid as a white solid, mp 129-131°.

20

M. [1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-[4-[(Pentyl-amino)carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl]methyl]- α , α -dimethylbenzenepropanoic acid, methyl ester

25

To a solution of 300 mg (0.73 mmol) of Part L oxazole acid in 3 mL of methylene chloride (distilled from phosphorous pentoxide) was added at room temperature a small drop of DMF then 85 μ L (0.97 mmol, Aldrich) of oxalyl chloride (gas evolution). The solution was stirred for 20 minutes then concentrated in vacuo to give the crude acid chloride as a foam.

30

To a solution of the crude acid chloride (~0.73 mmol) in 2 mL of dry methylene chloride cooled in an ice-bath was added dropwise a solution of 87 mg (1.0 mmol, Aldrich) of n-amylamine and 101 mg (1.0 mmol, distilled from calcium hydride) of triethylamine in 2 mL of methylene chloride. The reaction mixture was stirred for 10 minutes then partitioned between 20 mL of ethyl acetate and 20 mL of 1M aqueous HCl solution. The organic layer was separated, washed with 20 mL of brine, dried (magnesium sulfate) and concentrated in vacuo to give a solid. The crude solid was purified by flash chromatography (Merck silica, 15x3.0 cm, 2:1 ethyl acetate/hexane) to afford 331 mg (0.69 mmol, 94%) of title compound as a white solid, mp 130-131°.

Example 2

[1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-[4-[(Pentylamino)carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl)methyl]- α,α -dimethylbenzenepropanoic acid

A mixture of 280 mg (0.58 mmol) of Example 1 methyl ester and 840 mg (20 mmol, Aldrich) of lithium hydroxide monohydrate in 10 mL of 1:1 p-dioxane/water was stirred rapidly at 55° for 3 h. The reaction mixture was cooled in an ice-bath and acidified by dropwise addition of 2 mL of concentrated HCl then partitioned between 20 mL of water and 20 mL of ethyl acetate. The aqueous layer was separated and extracted with an additional 20 mL of ethyl acetate. The combined organic layers were dried (magnesium sulfate) and concentrated in vacuo to give a solid. The crude solid was recrystallized

(ethylacetate/hexane) to afford 225 mg (0.40 mmol, 83%) of title acid as a white solid, mp 139-140°.

IR(KBr): 3400 (broad), 2959, 2934, 1717, 1638,
5 1603, 1526 cm^{-1} .

MS(CI): 469 (M+H)⁺.

OR: $[\alpha]_D^{25} = +33^\circ$ (c=0.25 in chloroform).

10

TLC: R_f (silica gel, 1:9 methanol/methylene chloride)=0.58, ammonium molybdate/ceric sulfate and UV, homogeneous.

15 Analysis Calc'd for $\text{C}_{27}\text{H}_{36}\text{N}_2\text{O}_5$:

C, 69.21; H, 7.74; N, 5.98

Found: C, 69.31; H, 7.64; N, 6.07.

Example 3

20 [1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-[4-[(4-Cyclohexylbutyl)-carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl)methyl]- α,α -dimethylbenzenepropanoic acid,
methyl ester

25 To a solution of 89 mg (0.21 mmol) of Example 1, Part L oxazole acid in 2 mL of methylene chloride (distilled from phosphorous pentoxide) at room temperature was added a small drop of DMF then 25 μL (0.28 mmol, Aldrich) of oxalyl chloride. The
30 reaction mixture was stirred until gas evolution ceased (~15 minutes) then concentrated in vacuo to give the crude acid chloride as a pale yellow oil.

To a solution of the acid chloride (~0.21 mmol) in 2 mL of dry methylene chloride cooled in an ice-bath, was added dropwise a solution of 39 mg (0.25 mmol) of 4-cyclohexylbutylamine and 30 mg (0.30 mmol, distilled from calcium hydride) of triethylamine in 1 mL of dry methylene chloride. The reaction mixture was stirred for 10 minutes then partitioned between 15 mL of 1M aqueous HCl solution and 15 mL of ethyl acetate. The aqueous layer was separated and extracted with an additional 15 mL of ethyl acetate. The organic layers were combined, dried (magnesium sulfate) and concentrated in vacuo to give a solid. The crude material was purified by flash chromatography (Merck silica, 12x1.5 cm, 2:1 ethyl acetate/hexane) to afford 115 mg (0.21 mmol, 100%) of title ester as a white solid, mp 148-149°.

Example 4

[1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-[4-[(4-Cyclohexylbutyl)-amino]carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]-hept-2-yl]methyl]- α,α -dimethylbenzenepropanoic acid

A mixture of 110 mg (0.20 mmol) of Example 3 ester and 630 mg (15 mmol, Aldrich) of lithium hydroxide in 7.5 mL of 2:1 p-dioxane/water was stirred rapidly at 55° for 5 h. The reaction mixture was cooled in an ice-bath, acidified with 20 mL of 1M aqueous HCl solution then extracted with two-20 mL portions of ethyl acetate. The organic extracts were combined, dried (magnesium sulfate) and concentrated in vacuo to give a solid. The crude material was

recrystallized (hot acetonitrile) to afford 80 mg (0.15 mmol, 75%) of title acid as a white solid, mp 133-135°C.

5

Example 5

[1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-[4-[(4-Cyclohexylbutyl)-amino]carbonyl]-1H-imidazol-2-yl]-7-oxabicyclo-[2.2.1]hept-2-yl]methyl]- α , α -dimethylbenzene-propanoic acid, methyl ester

10

A. 3-Amino-2-[[[(1,1-dimethylethoxy)-carbonyl]amino]propanoic acid, benzyl ester

15

To a stirred mixture of [bis(trifluoro-acetoxy)iodosylbenzene (2.00 g, 4.66 mmol) in 24 mL of 1:1 DMF-water was added N- α -Boc-asparagine benzyl ester (1.00 g, 3.11 mmol, preparation was described by Wang, G. et al, in J. Org. Chem., Vol, 42, p 1286-1290, 1977). This mixture was stirred in a cold water bath for 15 minutes at which time dry pyridine (0.50 mL, 6.21 mmol) was added. The mixture was stirred at room temperature for 4 hours and concentrated in vacuo. The crude product was partitioned between 10 mL of 1N HCl solution and ether (4 X 15 mL). The aqueous layer was neutralized with NaHCO₃, saturated with NaCl and extracted with EtOAc (4 X 15 mL). The combined EtOAc extracts were dried (MgSO₄), filtered and concentrated in vacuo to give 0.53 g (58%) of title amine.

30

TLC: silica gel, 6% CH₃OH/CH₂Cl₂, R_f 0.44, Ce(SO₄)₂.

B. [1S-[1 α ,2 α ,3 α ,4 α]]-2-[[3-[[[2-[[[1,1-Dimethylethoxy)carbonyl]amino]-3-oxo-3-(phenylmethoxy)propyl]amino]oxomethyl]-7-oxabicyclo[2.2.1]hept-2-yl]methyl]- α,α -dimethylbenzenepropanoic acid, methyl ester

To a stirred mixture of Example 1, Part H acid (11.8 mmol), 1-hydroxybenzotriazole monohydrate (11.8 mmol) and Part A amine (11.8 mmol) in dry DMF under argon at 0°C is added sequentially (C₂H₅)₃N (23.6 mmol) and ethyl-3-(3-dimethylamino)propyl carbodiimide hydrochloride salt (11.8 mmol). The mixture is stirred at room temperature for 12 hours and concentrated in vacuo. The crude product is diluted with EtOAc and washed with 0.1N NaOH solution, 1N HCl solution, saturated NaHCO₃ solution and brine. The EtOAc layer is dried (MgSO₄), filtered and concentrated in vacuo. This is chromatographed on Merck silica gel 60 to give title amide.

C. [1S-[1 α ,2 α ,3 α ,4 α]]-2-[[3-[[[2-[[[1,1-Dimethylethoxy)carbonyl]amino]-3-oxo-3-(phenylmethoxy)propyl]amino]thioxomethyl]-7-oxabicyclo[2.2.1]hept-2-yl]methyl]- α,α -dimethylbenzenepropanoic acid, methyl ester

To a stirred mixture of Part B amide (1.12 mmol) in 14 mL benzene under argon is added Lawesson's reagent (0.72 mmol). The mixture is heated at 65°C under argon for 2 hours and cooled to room temperature. The mixture is diluted with ether and washed with saturated NaHCO₃ solution and brine. The organic layer is dried (MgSO₄), filtered

and concentrated in vacuo. Purification is effected by flash chromatography on Merck silica gel 60 to give title thioamide.

5 D. [1S-[1 α ,2 α ,3 α ,4 α]]-2-[[3-[1-[(1,1-Dimethylethoxy)carbonyl]-4,5-dihydro-5-[(phenylmethoxy)carbonyl]-1H-imidazol-2-yl]-7-oxabicyclo[2.2.1]hept-2-yl]-methyl]- α , α -dimethylbenzenepropanoic acid,
10 methyl ester

To a stirred mixture of Part C thioamide (0.57 mmol), (C₆H₅)₃P (1.71 mmol) and (C₂H₅)₃N (1.71 mmol) in acetonitrile is added CCl₄ (6.27 mmol). The mixture is stirred at room temperature for 4
15 hours and diluted with ether and water. The resulting mixture is saturated with NaCl and extracted with ether. The combined ether extracts are dried (MgSO₄), filtered and concentrated in vacuo. This is chromatographed on Merck silica gel 60 to give
20 title Boc (or BOC)-imidazoline.

E. [1S-[1 α ,2 α ,3 α ,4 α]]-2-[[3-[5-[[4-Cyclohexylbutyl)amino]carbonyl]-[1-[(1,1-dimethylethoxy)carbonyl]-4,5-
25 dihydro-1H-imidazol-2-yl]-7-oxabicyclo[2.2.1]hept-2-yl]methyl]- α , α -dimethyl-benzenepropanoic acid, methyl ester

To a stirred mixture of Part D Boc-imidazoline (0.32 mmol) in methanol under argon is added 20%
30 Pd/C (20% based on the weight of Part D compound). The atmosphere is replaced with hydrogen by several vacuum-fill cycles. The mixture is stirred at room temperature for 4.5 hours and the catalyst is

filtered off through a 0.4 μ m polycarbonate film. The catalyst is rinsed with DMF. The filtrate is concentrated in vacuo to give crude acid. To a stirred mixture of this acid, 1-hydroxybenzotriazole monohydrate (0.32 mmol) and 4-cyclohexylbutyl amine hydrochloride salt (0.38 mmol) in DMF under argon at 0°C is added sequentially $(C_2H_5)_3N$ (0.79 mmol) and ethyl-3-(3-dimethylamino)propyl carbodiimide hydrochloride salt (0.32 mmol). The mixture is stirred at room temperature for 18 hours and concentrated in vacuo. The crude product is partitioned between EtOAc and 0.1N NaOH solution, 1N HCl solution and saturated $NaHCO_3$ solution. The organic layer is dried ($MgSO_4$), filtered and concentrated in vacuo. This is chromatographed on Merck silica gel 60 to give title amide.

F. [1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-[4-[[[4-Cyclohexylbutyl)amino]carbonyl]-1H-imidazol-2-yl]-7-oxabicyclo[2.2.1]hept-2-yl)methyl]- α,α -dimethylbenzenepropanoic acid, methyl ester

To a stirred mixture of Part E amide (0.94 mmol) in of dry CH_2Cl_2 at 0°C is added trifluoroacetic acid (TFA). The mixture is stirred at room temperature for 3 hours. The mixture is diluted with 40 mL of toluene and concentrated in vacuo. The crude imidazole-TFA salt is diluted with EtOAc and washed once with saturated $NaHCO_3$ solution. The aqueous layer is extracted with EtOAc. The combined EtOAc extracts are dried ($MgSO_4$), filtered and concentrated in vacuo. To this crude imidazoline in $CHCl_3$ is added MnO_2 (6.55 mmol). The mixture is

stirred at room temperature for 64 hours at which
time MnO_2 (6.55 mmol) is added. The mixture is
stirred at room temperature for 1 day and another
amount of MnO_2 (3.28 mmol) is added. The mixture
5 is stirred at room temperature for one more day
and again MnO_2 (2.18 mmol) is added. The mixture
is stirred at room temperature for 1 day and MnO_2
is filtered off through a pad of Celite and the pad
is rinsed with CHCl_3 . The filtrate is concentrated
10 in vacuo and chromatographed on Merck silica gel 60
to give title imidazole.

Example 6

15 [1S-(1 α , 2 α , 3 α , 4 α)]-2-[[3-[4-[(4-Cyclohexylbutyl)-
amino]carbonyl]-1H-imidazol-2-yl]-7-oxabicyclo-
[2.2.1]hept-2-yl]methyl]- α , α -dimethylbenzenepropa-
noic acid, hydrochloride salt

To a stirred mixture of Example 5 imidazole
20 (0.05 mmol) in 1 mL of methanol is added 2N KOH
solution. The mixture is stirred at room temperature
for 4 hours and concentrated in vacuo to remove
methanol. The residue is diluted with CH_2Cl_2 and
acidified to pH 2 by the addition of 1N HCl solution.
25 The aqueous layer is separated and extracted with
 CH_2Cl_2 . The combined CH_2Cl_2 extracts are dried
(Na_2SO_4), filtered and concentrated in vacuo. The
crude product is dissolved in CH_2Cl_2 and combined
with 4N HCl in ether. The resulting mixture is
30 concentrated in vacuo and triturated in hot EtOAc.
The mixture is cooled to room temperature and the
solid formed is collected by filtration to give
title hydrochloride salt.

Example 7

[1S-[1 α ,2 α (Z),3 α ,4 α]]-6-[3-[4-[[4-Cyclohexylbutyl)-amino]carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl]-4- α , α -dimethylhexenoic acid

5

A. [(1,1-Dimethylethoxy)carbonyl]-N-(4-cyclohexylbutyl)-L-serinamide

To a solution of 575 mg of 4-cyclohexylbutyl-amine hydrochloride (3.0 mmol), 615 mg t-butyloxy-carbonyl-L-serine (3.0 mmol, 1.0 equiv), 405 mg 10 1-hydroxybenzotriazole hydrate (3.0 mmol, 1.0 equiv), and 387 mg diisopropylethylamine (3.0 mmol, 1.0 equiv) in 10 mL dry tetrahydrofuran (THF) stirring under argon at 0°, was added 618 mg 1,3-dicyclohexyl-carbodiimide (3.0 mmol, 1.0 equiv) in a single 15 portion. A precipitate slowly formed. After 1 hour the mixture was warmed to room temperature and stirred for 4 hours. After dilution with ethyl acetate, the mixture was filtered, and the filtrate 20 was washed with a pH 1 salt solution (formed by mixing water, brine, and 1 M aqueous HCl solution). Further washing (twice) with 1 M NaHCO₃ was followed by drying over Na₂SO₄ and evaporation to give 1.1 g of crude title amide.

25

B. N-(4-Cyclohexylbutyl)-L-serinamide

To a solution of 1.1 g crude Part A amide in 4 mL CH₂Cl₂ at room temperature was added 4 mL trifluoroacetic acid. The mixture was stirred for 30 4 hours. After solvent evaporation, residual trifluoroacetic acid was azeotropically removed by rotoevaporation with CHCl₃. Flash chromatography (150 g silica, 10% [10% concentrated aqueous NH₃

in CH₃OH] in CH₂Cl₂) gave, after azeotroping with
toluene and exposure to high vacuum, 495 mg of
pure title amine as a white solid. The yield of
title amine was 68% overall from 4-cyclohexylbutyl
5 amine hydrochloride.

C. [1S-[1 α ,2 α (Z),3 α ,4 α]]-6-[3-(Hydroxy-
methyl)-7-oxabicyclo[2.2.1]hept-2-yl]-4-
 α , α -dimethylhexenoic acid, methyl ester

10 To a partial solution of [4aR-
(4 $\alpha\alpha$,5 β ,8 β ,8 $\alpha\beta$)]-octahydro-5,8-epoxy-1H-2-
benzopyran-3-ol (prepared as described in U.S.
Patent No. 4,143,054) (23 mmol) and α , α -dimethyl-
3-carboxypropyltriphenylphosphonium bromide (37
15 mmol) in dry THF under argon at 3°C is added
dropwise over 1 hour a solution of of potassium
t-amylate (68 mmol of a 1.8M toluene solution) with
mechanical stirring. The reaction is then run at
room temperature for 90 minutes. A 0°C ice bath
20 is introduced and the reaction is quenched by the
addition of glacial acetic acid, over 30 minutes.
Solvents are removed in vacuo (azeotroped with
toluene). Water and concentrated HCl are added
(pH 2.6).

25 The mixture is extracted with ethyl acetate,
dried (sodium sulfate) and concentrated in vacuo.
The crude material is dissolved in methanolic HCl,
stirred for 24 h and concentrated in vacuo. The
crude material is purified by flash chromatography
30 (Merck silica) to give title methyl ester.

D. [1S-[1 α ,2 α (Z),3 α ,4 α]]-6-[3-(Carboxy)-
7-oxabicyclo[2.2.1]hept-2-yl]-4- α , α -di-
methylhexenoic acid, methyl ester

To a solution of Part C alcohol (7.6 mmol)
5 in acetone under argon at 0°, is added slowly Jones'
Reagent (2.6 M in Cr^{VI}). The resulting mixture is
stirred for 20 minutes before 2-propanol is added to
quench excess reagent. Still at 0°, 3 M aqueous
NaHSO₃ solution is added with stirring until all
10 salts dissolved. Brine was added, and extraction
with ethyl acetate followed. After drying the
extracts over Na₂SO₄, solvent evaporation, and flash
chromatography (silica) afforded title acid.

15 E. [1S-[1 α ,2 α (Z),3 α (R*),4 α]]-6-[3-[[[2-
[(4-Cyclohexylbutyl)amino]-1-(hydroxymeth-
yl)-2-oxoethyl]amino]carbonyl]-7-oxabicyclo-
[2.2.1]hept-2-yl]-4- α , α -dimethylhexenoic
acid, methyl ester

20 To a solution of Part D acid in DMF cooled
in an ice-bath is added 1-hydroxybenzotriazole (2.4
mmol), triethylamine (2.4 mmol) and Part B amine
(2.4 mmol) then after several minutes WSC (2.4 mmol).
The reaction mixture is stirred at 0° for 2 h then
25 at 25° for 16 h, partitioned between ethyl acetate
and 1M HCl solution. The organic layer is washed
with 1M HCl, 1M NaOH, brine, dried (magnesium
sulfate) and concentrated in vacuo. Purification
by flash chromatography (Merck silica) afforded
30 title amide.

F. [1S-[1 α ,2 α (Z),3 α (R*),4 α]]-6-[3-[4-[(4-Cyclohexylbutyl)amino]carbonyl]-4,5-dihydro-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl]- α,α -dimethyl-4-hexenoic acid, methyl ester

5 To a solution of pure Part E hydroxybisamide (0.48 mmol) in dry acetonitrile under argon at room temperature, is added triphenylphosphine (0.72 mmol, 1.5 equiv), triethylamine (0.72 mmol, 1.5 equiv), and CCl₄ (0.58 mmol, 1.2 equiv), and the mixture is
10 stirred at room temperature for 18 h. Solvent evaporation followed by flash chromatography (silica) affords pure title oxazoline.

15 G. [1S-[1 α ,2 α (Z),3 α ,4 α]]-6-[3-[4-[(4-Cyclohexylbutyl)amino]carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl]-4- α,α -di-methylhexenoic acid, methyl ester

To a solution of pure Part F oxazoline (0.40 mmol) in CH₂Cl₂ is added 200 mg NiO₂, and the
20 heterogenous mixture is stirred at room temperature. Over 1 day, five additional aliquots of the reagent are added until the reaction is complete. The mixture is diluted with ethyl acetate, and this is stirred with 3 M aqueous NaHSO₃ solution until the
25 black color of the NiO₂ disappeared and most of the solids dissolved. Extraction with ethyl acetate is followed by drying over Na₂SO₄ and evaporation. Flash chromatography (silica) affords pure title
30 oxazole.

H. [1S-[1 α ,2 α (Z),3 α ,4 α]]-6-[3-[4-[[4-(Cyclohexylbutyl)amino]carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl]-4- α , α -dimethylhexenoic acid

- 5 To pure Part G oxazole (0.19 mmol) in CH₃OH at room temperature, is added 1.0 M aqueous NaOH solution. After stirring the mixture for 1.3 hours, 1 M aqueous HCl solution is added to lower the pH to 1. Extraction with ethyl acetate followed.
- 10 The extracts are dried over Na₂SO₄, and solvent evaporation gives crude title acid. Flash chromatography (Merck silica) affords pure title acid.

Example 8

- 15 [1S-(1 α ,2 α ,3 α ,4 α)]-3-[4-[[4-(Cyclohexylbutyl)-amino]carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]-heptane-2- α , α -dimethylhexanoic acid

- 20 A solution of Example 7 acid in ethyl acetate and acetic acid is degassed via a vacuum-fill cycle with argon. To this solution is added 10% Pd/C and the atmosphere is exchanged for hydrogen by two vacuum-fill cycles. A slight positive pressure is maintained through use of a hydrogen balloon. The
- 25 mixture is stirred at room temperature for 22.5 hours, diluted with CH₂Cl₂ and filtered through a polycarbonate filter to remove the catalyst. The filtrate is concentrated in vacuo to afford pure title acid.

Example 9

[1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-[4-[(Heptylamino)carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl)methyl]-benzene- α , α -dimethylpropanoic acid, methyl ester

5

To a solution of acid prepared in Example 1, Part L (0.52 mmol) in dry CH₂Cl₂ (distilled from P₂O₅) is added 1 small drop of DMF, followed by (0.63 mmol) of oxalyl chloride. The reaction is stirred until gas evolution ceased (about 30 minutes), then the mixture is concentrated in vacuo to give the crude acid chloride as a pale yellow solid.

To a solution of crude acid chloride in dry CH₂Cl₂ (distilled from P₂O₅), cooled to 0°, is added (0.83 mmol) triethylamine, followed by the dropwise addition of a solution (0.62 mmol) of heptylamine in CH₂Cl₂. The reaction is stirred at 0° for 1.5 hours, then partitioned between ethyl acetate/1M HCl. The ethyl acetate layer is separated, dried (MgSO₄) and concentrated in vacuo to give a crude orange solid. The crude solid is flash chromatographed (Merck silica) to give title ester.

25

Example 10

[1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-[4-[(Heptylamino)carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl)methyl]-benzene- α , α -dimethylpropanoic acid

30

To a mixture of (0.37 mmol) of Example 9 ester in THF/water is added (0.75 mmol, Aldrich) lithium hydroxide monohydrate. The reaction is stirred vigorously at room temperature for 3 hours, then

quenched by the addition of 1M HCl. The mixture is partitioned between ethyl acetate/water. The ethyl acetate layer is separated, dried (MgSO_4) and concentrated in vacuo to give a crude white solid.
5 The crude solid is recrystallized to give title acid.

Example 11

[1S-[1 α ,2 α (Z),3 α ,4 α]]-6-[3-[5-[[4-Cyclohexylbutyl)-amino]carbonyl]-1H-imidazol-2-yl]-7-oxabicyclo-
10 [2.2.1]hept-2-yl]-4- α , α -dimethylhexenoic acid, methyl ester

The title compound may be prepared employing the procedures set out in Example 5 except that the
15 Example 7, Part D acid is employed in place of the Example 1, Part H acid.

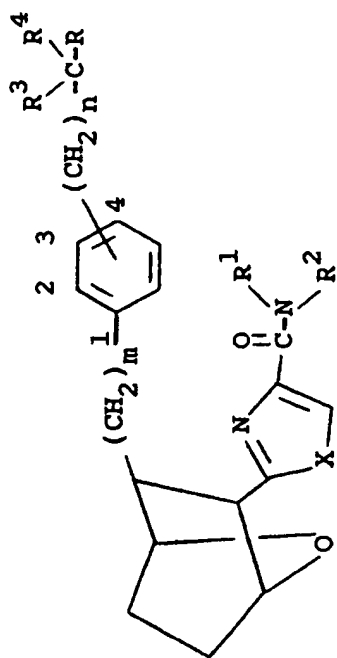
Example 12




[1S-[1 α ,2 α (Z),3 α ,4 α]]-6-[3-[4-[[4-Cyclohexylbutyl)-amino]carbonyl]-1H-imidazol-2-yl]-7-oxabicyclo-
20 [2.2.1]hept-2-yl]-4- α , α -dimethylhexenoic acid


To a solution of Example 11 ester in methanol is added 2N KOH. The reaction is stirred at room
25 temperature for 4 hours. The methanol is removed in vacuo. The residue is taken up in methylene chloride and brought to pH 2 with 1N HCl. After shaking, the aqueous layer is further extracted with methylene chloride. The combined organic layers are dried
30 (Na_2SO_4) and concentrated in vacuo. The residue is taken up in methylene chloride and ethereal HCl was added. The mixture is concentrated in vacuo. Trituration with of ethyl acetate yields a white

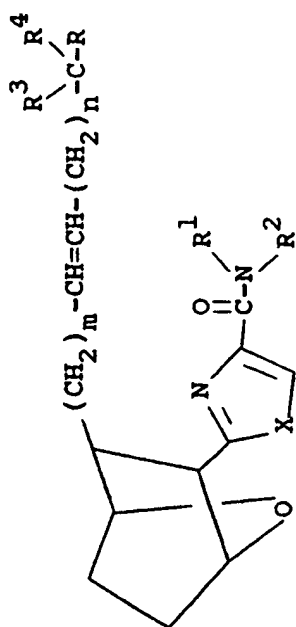
solid which is collected by filtration and dried to yield title acid.

- 5 Examples of additional compounds in accordance with the present invention which may be prepared following the procedures outlined in the specification and working Examples include, but are not limited to the following.




Example No.	$\frac{(\text{CH}_2)_m}{m}$	$\frac{(\text{CH}_2)_n}{n}$	$\frac{-(\text{CH}_2)_n}{(\text{position})}$	$\frac{X}{R^1}$	$\frac{R^2}{R}$	$\frac{R^3}{R}$	$\frac{R^4}{R}$
13	1	2	-(2)	0	$-\text{C}_6\text{H}_{13}$	CH_3	CH_3
14	2	2	-(2)	0	$-(\text{CH}_2)_4$ - 	CH_3	C_2H_5
15	2	3	-(2)	0	C_6H_5	C_6H_5	CH_3
16	1	2	-(3)	NH	$-(\text{CH}_2)_3$ - 	CO_2H	CH_3
17	2	2	-(3)	0		CO_2CH_3	C_2H_5
					$\text{CH}_2\text{C}_6\text{H}_5$	C_2H_5	CH_3

Example No.	$\frac{(\text{CH}_2)_m}{m}$	$\frac{(\text{CH}_2)_n}{n}$	$\frac{-(\text{CH}_2)_n}{(\text{position})}$	$\underline{\text{X}}$	$\frac{\text{R}^1}{\text{---}}$	$\frac{\text{R}^2}{\text{---}}$	$\frac{\text{R}}{\text{---}}$	$\frac{\text{R}^3}{\text{---}}$	$\frac{\text{R}^4}{\text{---}}$
18	1	2	-(2)	O		H	$\text{CO}_2\text{C}_2\text{H}_5$	C_2H_5	C_2H_5
19	1	2	-(2)	O	$(\text{CH}_2)_2\text{C}_6\text{H}_5$	CH_3	CO_2H	$\underbrace{\text{---CH}_2\text{CH}_2\text{---}}_{\text{---CH}_2\text{CH}_2\text{---}}$	
20	1	2	-(3)	NH	$n\text{-C}_5\text{H}_{11}$	H	CO_2H	C_2H_5	CH_3
21	1	2	-(2)	O	$\underbrace{\text{---}(\text{CH}_2)_6\text{---}}_{\text{---}(\text{CH}_2)_6\text{---}}$		CO_2H	CH_3	CH_3
22	2	0	-(3)	O	$n\text{-C}_4\text{H}_9$	$n\text{-C}_4\text{H}_9$	CO_2H	CH_3	CH_3
23	1	2	-(2)	O	$\text{---}(\text{CH}_2)_4\text{---N} \begin{array}{c} \diagup \text{N} \\ \diagdown \end{array} \text{N} \text{---}$	H	CO_2H	C_4H_9	C_4H_9
24	2	2	-(3)	O	$\text{---}(\text{CH}_2)_5\text{---} \begin{array}{c} \text{C}_6\text{H}_4 \\ \text{N} \end{array} \text{---}$	H	CO_2H	CH_3	C_2H_5
25	1	2	-(2)	O	$\text{---}(\text{CH}_2)_4\text{---} \begin{array}{c} \text{C}_6\text{H}_8\text{O} \end{array} \text{---}$	H	CO_2H	CH_3	CH_3

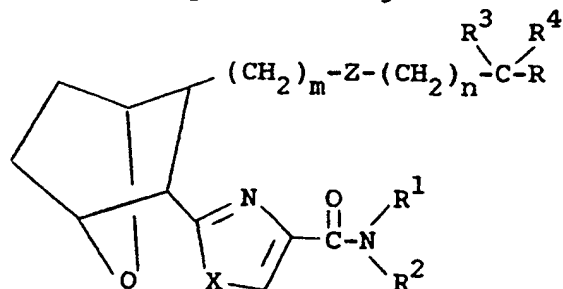


Example No.	$\frac{(\text{CH}_2)_m}{m}$	$\frac{(\text{CH}_2)_n}{n}$	X	$\frac{R^1}{\text{---}}$	$\frac{R^2}{\text{---}}$	$\frac{R}{\text{---}}$	$\frac{R^3}{\text{---}}$	$\frac{R^4}{\text{---}}$
26	1	2	O	C_6H_{13}	CH_3	CO_2H	CH_3	CH_3
27	2	2	O	$-(\text{CH}_2)_2-\text{S}$	C_2H_5	CO_2H	C_2H_5	C_2H_5
28	2	3	O	C_6H_5	C_6H_5	CO_2H	CH_3	C_2H_5
29	2	2	O		$\text{CH}_2\text{C}_6\text{H}_5$	CO_2CH_3	CH_3	CH_3

Example No.	$\frac{(\text{CH}_2)_m}{m}$	$\frac{(\text{CH}_2)_n}{n}$	$\frac{X}{X}$	$\frac{R^1}{R^1}$	$\frac{R^2}{R^2}$	$\frac{R}{R}$	$\frac{R^3}{R^3}$	$\frac{R^4}{R^4}$
30	1	2	NH	C ₂ H ₅	H	CO ₂ Li	C ₂ H ₅	CH ₃
31	1	3	NH		C ₂ H ₅	CO ₂ C ₂ H ₅	C ₂ H ₅	C ₂ H ₅
32	1	2	O	-(CH ₂) ₂ C ₆ H ₅	CH ₃	CO ₂ H	CH ₃	CH ₃
33	1	2	O	$\underbrace{-(\text{CH}_2)_6}_{6^-}$		CO ₂ H	CH ₃	CH ₃

What we claim is:

1. A compound having the formula

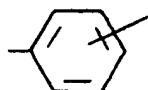


including all stereoisomers thereof, wherein

m is 1, 2 or 3;

n is 0, 1, 2, 3 or 4;

Z is $-(CH_2)_2-$, $-CH=CH-$ or



with the proviso that when Z is $-CH=CH-$, n is 1, 2, 3 or 4;

R is CO_2H , CO_2 alkali metal, or CO_2 lower alkyl;

X is O or NH;

R^1 is hydrogen, lower alkyl, lower alkenyl, lower alkynyl, aryl, aralkyl, cycloalkyl, cycloalkylalkyl, cycloheteroalkyl, cycloheteroalkylalkyl, heteroaryl or heteroarylalkyl, or an amide

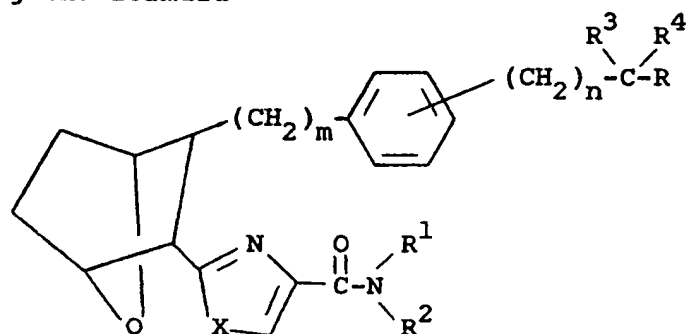
$(-(CH_2)_t-\overset{\overset{O}{\parallel}}{C}-\overset{\overset{H}{\mid}}{N}-Ra$ or $-(CH_2)_t-\overset{\overset{H}{\mid}}{N}-\overset{\overset{O}{\parallel}}{C}-Ra$ wherein t is 1 to 12 and Ra is lower alkyl, aryl, cycloalkyl or cycloalkylalkyl); each of R^1 being unsubstituted or optionally substituted with a lower alkyl, aryl, cycloalkyl or cycloalkylalkyl group;

R^2 is hydrogen, lower alkyl, aryl, or aralkyl, or

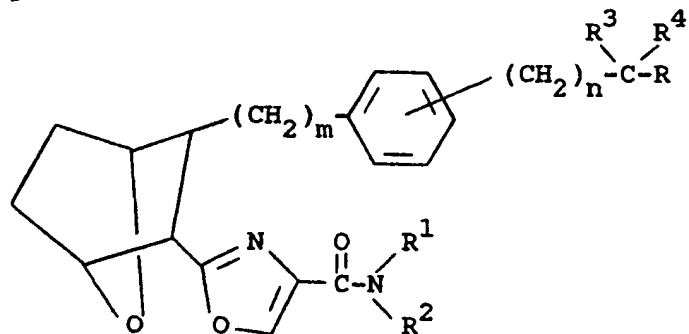
R^1 and R^2 together with the N to which they are linked form a 5- to 8- membered ring; and

R^3 and R^4 are the same or different and each is lower alkyl, R^3 and R^4 may be linked to form a 3- or 4-membered ring, including pharmaceutically acceptable salts thereof.

2. The compound as defined in Claim 1 having the formula

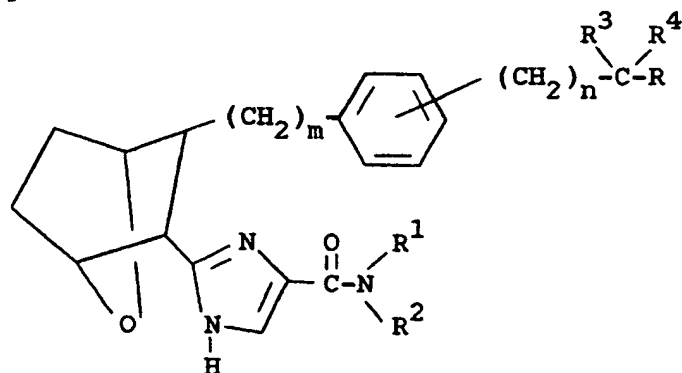


3. The compound as defined in Claim 2 having the formula



4. The compound as defined in Claim 2 where $m=1$ and $n=2$.

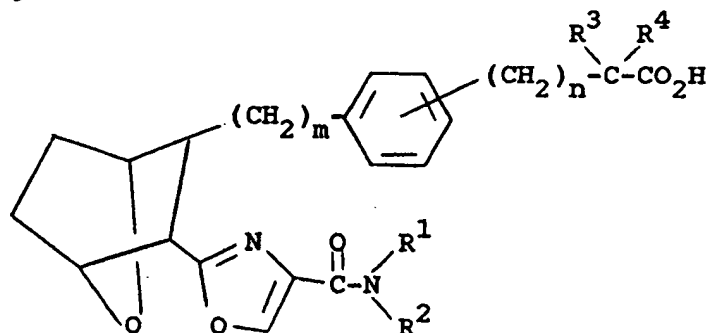
5. The compound as defined in Claim 2 having the formula



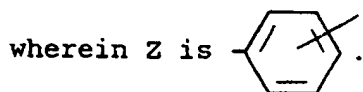
6. The compound as defined in Claim 1 wherein R^3 and R^4 are each the same alkyl.

7. The compound as defined in Claim 1 wherein R^3 and R^4 are each methyl.

8. The compound as defined in Claim 3 having the formula



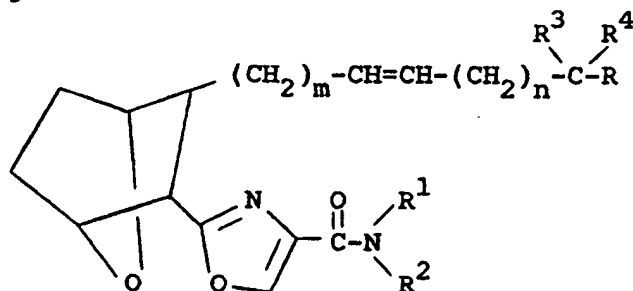
9. The compound as defined in Claim 1



10. The compound as defined in Claim 1
wherein Z is $-(CH_2)_2-$ or $-CH=CH-$.

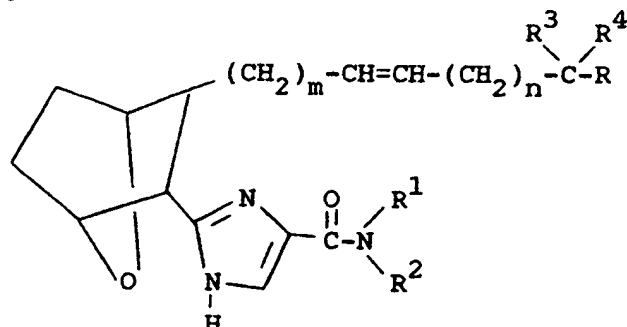
11. The compound as defined in Claim 2
having the name [1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-[4-[(pent-ylamino)carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]-hept-2-yl]methyl]- α,α -dimethylbenzenepropanoic acid, or esters or salts thereof; or [1S-(1 α ,2 α ,3 α ,4 α)]-2-[[3-[4-[[4-cyclohexylbutyl]amino]carbonyl]-2-oxazolyl]-7-oxabicyclo[2.2.1]hept-2-yl]methyl]- α,α -dimethylbenzenepropanoic acid, or esters, or salts thereof.

12. The compound as defined in Claim 1
having the formula



13. The compound as defined in Claim 12
where $m=1$ and $n=2$.

14. The compound as defined in Claim 12 having the formula



15. The compound as defined in Claim 12 wherein R^3 and R^4 are each the same lower alkyl.

16. The compound as defined in Claim 12 wherein R^3 and R^4 are each methyl.

17. A method of inhibiting platelet aggregation and bronchoconstriction, which comprises administering to the circulatory system of a mammalian host an effective amount of a compound as defined in claim 1.

18. The method as defined in Claim 17 wherein said compound is administered in an amount within the range of from about 0.1 to about 100 mg/kg.

19. A composition for inhibiting platelet aggregation and bronchoconstriction comprising an effective amount of a compound as defined in Claim 1, and a pharmaceutically acceptable carrier therefor.

20. A method of inhibiting platelet aggregation which comprises administering to a mammalian host an effective amount of a compound as defined in Claim 1.

21. A method of inhibiting bronchoconstriction associated with asthma, which comprises administering to a mammalian host an effective amount of a compound as defined in Claim 1.

22. A method for improving post-ischemic myocardial function, which comprises administering to a mammalian host in need of such treatment an effective amount of a compound as defined in Claim 1.

23. A method for treating toxemia during pregnancy, which comprises administering to a mammalian host in need of such treatment an effective amount of a compound as defined in Claim 1.

24. A method for preventing or reducing venous thrombosis, which comprises administering to a mammalian host in need of such treatment an effective amount of a compound as defined in Claim 1.

25. A method for preventing or reducing platelet loss during extracorporeal circulation, which comprises administering to a mammalian host in need of such treatment an effective amount of a compound as defined in Claim 1.

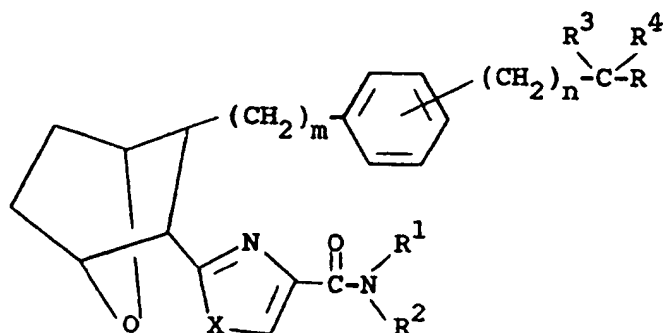
26. A method for treating burn injuries and/or promoting wound healing, which comprises administering to a mammalian host in need of such treatment an effective amount of a compound as defined in Claim 1 in systemic or topical form.

27. A method for reducing post-ischemic myocardial injury, which comprises administering to a mammalian host in need of such treatment an

effective amount of a compound as defined in Claim 1 and an effective amount of a thrombolytic agent within 6 hours of a myocardial infarction.

28. The method as defined in Claim 31 wherein said thrombolytic is t-PA, streptokinase, urokinase, prourokinase or anisoylated plasminogen-streptokinase activator complex.

29. A compound having the formula



including all stereoisomers thereof, wherein

m is 1, 2 or 3;

n is 0, 1, 2, 3 or 4;

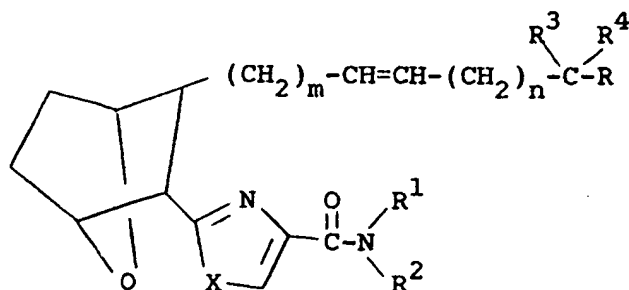
R is CO_2H , CO_2 alkali metal or CO_2 lower alkyl;

X is O or NH;

R^1 is lower alkyl, aryl, aralkyl, cycloalkyl, cycloalkylalkyl, saturated heterocycle, or aromatic heterocycle; and

R^2 is hydrogen, lower alkyl, aryl, or aralkyl, or R^1 and R^2 together with the N to which they are linked form a 5- to 8- membered ring; and R^3 and R^4 are the same or different and each is lower alkyl; R^3 and R^4 may be linked to form a 3- or 4-membered ring, including pharmaceutically acceptable salts thereof.

30. A compound having the formula



including all stereoisomers thereof, wherein

m is 1, 2 or 3;

n is 1, 2, 3 or 4;

R is CO₂H, CO₂alkali metal or CO₂lower alkyl;

X is O or NH;

wherein R¹ is lower alkyl, aryl, aralkyl, cycloalkyl, or cycloalkylalkyl;

R² is hydrogen, lower alkyl, aryl, or aralkyl, or R¹ and R² together with the N to which they are

linked form a 5- to 8- membered ring; and R³ and R⁴

are the same or different and are each lower alkyl;

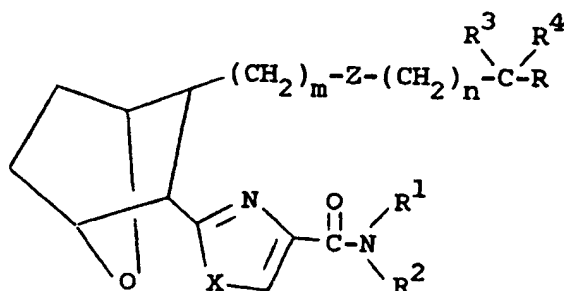
R³ and R⁴ may be linked to form a 3- or 4-membered ring, including pharmaceutically acceptable salts

thereof.

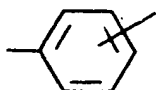
Abstract

GEM-DIALKYL-7-OXABICYCLOHEPTYL SUBSTITUTED
HETEROCYCLIC AMIDE PROSTAGLANDIN
ANALOGS USEFUL IN THE TREATMENT OF THROMBOTIC
AND VASOSPASTIC DISEASE

Gem-dialkyl-7-oxabicycloheptane substituted prostaglandin analogs useful in treating thrombotic and vasospastic disease have the structural formula



wherein m is 1, 2 or 3; n is 0, 1, 2, 3 or 4;
 Z is $-(CH_2)_2-$, $-CH=CH-$ or



with the proviso when Z is $-CH=CH-$, n is 1, 2, 3 or 4; R is CO_2H , CO_2 lower alkyl, or CO_2 alkali metal, X is O or NH; and where R^1 and R^2 are as defined herein and R^3 and R^4 are each independently alkyl, or R^3 and R^4 may be taken together with the carbon to which it is attached to form a 3- or 4-membered ring.